

# (2E)-Dodec-2-en-1-yl methyl ether

<b>Inchi:</b>	InChI=1S/C13H26O/c1-3-4-5-6-7-8-9-10-11-12-13-14-2/h11-12H,3-10,13H2,1-2H3/b12-1
<b>InchiKey:</b>	VHDLPJILDRILOW-VAWYXSNFSA-N
<b>Formula:</b>	C13H26O
<b>SMILES:</b>	CCCCCCCCC=CCOC
<b>Mol. weight [g/mol]:</b>	198.34

## Physical Properties

Property code	Value	Unit	Source
gf	33.80	kJ/mol	Joback Method
hf	-326.65	kJ/mol	Joback Method
hfus	30.82	kJ/mol	Joback Method
hvap	46.90	kJ/mol	Joback Method
log10ws	-4.21		Crippen Method
logp	4.330		Crippen Method
mcvol	195.600	ml/mol	McGowan Method
pc	1692.12	kPa	Joback Method
rinsol	1441.50		NIST Webbook
tb	523.42	K	Joback Method
tc	690.13	K	Joback Method
tf	253.42	K	Joback Method
vc	0.761	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	467.95	J/molxK	523.42	Joback Method
cpg	484.85	J/molxK	551.21	Joback Method
cpg	501.09	J/molxK	578.99	Joback Method
cpg	516.68	J/molxK	606.78	Joback Method
cpg	531.64	J/molxK	634.56	Joback Method
cpg	545.99	J/molxK	662.35	Joback Method
cpg	559.74	J/molxK	690.13	Joback Method
dvisc	0.0038192	Paxs	253.42	Joback Method
dvisc	0.0014455	Paxs	298.42	Joback Method

dvisc	0.0007057	Paxs	343.42	Joback Method
dvisc	0.0004068	Paxs	388.42	Joback Method
dvisc	0.0002629	Paxs	433.42	Joback Method
dvisc	0.0001845	Paxs	478.42	Joback Method
dvisc	0.0001376	Paxs	523.42	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U334063&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U334063&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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