

# Ethofumesate

<b>Other names:</b>	(.+/-)-2-Ethoxy-2,3-dihydro-3,3-dimethyl-5-benzofuranol methanesulfonate (.+/-)-2-Ethoxy-2,3-dihydro-3,3-dimethyl-5-benzofuranylmethanesulfonate (dl)-2-ethoxy-2,3-dihydro-3,3-dimethyl-5-benzofuranylmethanesulfonate 2-Ethoxy-3,3-dimethyl-2,3-dihydro-1-benzofuran-5-yl methanesulfonate 5-Benzofuranol, 2-Ethoxy-2,3-dihydro-3,3-dimethyl-, methanesulfonate 5-Benzofuranol, 2-ethoxy-2,3-dihydro-3,3-dimethyl-, methanesulfonate, (+/-)- CR 14658 NC 8438 Norton Nortran Nortranese Nortron Nortron (new) Prograss Tramat
<b>Inchi:</b>	InChI=1S/C13H18O5S/c1-5-16-12-13(2,3)10-8-9(18-19(4,14)15)6-7-11(10)17-12/h6-8,12
<b>InchiKey:</b>	IRCMYGHHKLLGHV-UHFFFAOYSA-N
<b>Formula:</b>	C13H18O5S
<b>SMILES:</b>	CCOC1Oc2ccc(OS(C)(=O)=O)cc2C1(C)C
<b>Mol. weight [g/mol]:</b>	286.34
<b>CAS:</b>	26225-79-6

## Physical Properties

Property code	Value	Unit	Source
gf	-565.38	kJ/mol	Joback Method
hf	-880.15	kJ/mol	Joback Method
hfus	37.33	kJ/mol	Joback Method
hvap	74.55	kJ/mol	Joback Method
log10ws	-3.42		Aqueous Solubility Prediction Method
log10ws	-3.42		Estimated Solubility Method
logp	2.058		Crippen Method
mcvol	205.110	ml/mol	McGowan Method
pc	2738.28	kPa	Joback Method
rinpol	1910.00		NIST Webbook
rinpol	1951.00		NIST Webbook

rinpol	1953.00		NIST Webbook
rinpol	1924.00		NIST Webbook
tb	655.36	K	Joback Method
tc	862.89	K	Joback Method
tf	345.62 ± 0.20	K	NIST Webbook
vc	0.792	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	552.27	J/mol×K	655.36	Joback Method
cpg	568.33	J/mol×K	689.95	Joback Method
cpg	583.58	J/mol×K	724.54	Joback Method
cpg	598.09	J/mol×K	759.13	Joback Method
cpg	611.93	J/mol×K	793.71	Joback Method
cpg	625.18	J/mol×K	828.30	Joback Method
cpg	637.92	J/mol×K	862.89	Joback Method
hfust	26.25	kJ/mol	344.10	NIST Webbook

## Sources

<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C26225796&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C26225796&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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