

# 6-Chrysenamine

<b>Other names:</b>	6-Amc 6-Aminochrysene Chrysenex Chrysonex chrysen-6-ylamine
<b>Inchi:</b>	InChI=1S/C18H13N/c19-18-11-17-13-6-2-1-5-12(13)9-10-15(17)14-7-3-4-8-16(14)18/h1-
<b>InchiKey:</b>	KIVUHCNVDWYUNP-UHFFFAOYSA-N
<b>Formula:</b>	C18H13N
<b>SMILES:</b>	<chem>Nc1cc2c3ccccc3ccc2c2ccccc12</chem>
<b>Mol. weight [g/mol]:</b>	243.30
<b>CAS:</b>	2642-98-0

## Physical Properties

Property code	Value	Unit	Source
gf	570.60	kJ/mol	Joback Method
hf	394.27	kJ/mol	Joback Method
hfus	31.50	kJ/mol	Joback Method
hvap	75.48	kJ/mol	Joback Method
ie	7.00 ± 0.10	eV	NIST Webbook
log10ws	-6.20		Aqueous Solubility Prediction Method
log10ws	-6.20		Estimated Solubility Method
logp	4.728		Crippen Method
mcvol	192.320	ml/mol	McGowan Method
pc	2865.80	kPa	Joback Method
rinpola	463.19		NIST Webbook
rinpola	452.52		NIST Webbook
tb	782.33	K	Joback Method
tc	1050.06	K	Joback Method
tf	483.65	K	Aqueous Solubility Prediction Method
vc	0.731	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	523.87	J/mol×K	782.33	Joback Method
cpg	537.53	J/mol×K	826.95	Joback Method
cpg	550.29	J/mol×K	871.57	Joback Method
cpg	562.36	J/mol×K	916.19	Joback Method
cpg	573.96	J/mol×K	960.81	Joback Method
cpg	585.32	J/mol×K	1005.44	Joback Method
cpg	596.64	J/mol×K	1050.06	Joback Method

## Sources

<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2642980&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2642980&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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