

# Urea, N-(2-chloroethyl)-N'-(2,6-dioxo-3-piperidiny)-N-nitro

<b>Other names:</b>	N-(2-Chloroethyl)-N'-(2,6-dioxo-3-piperidiny)-N-nitroso- NSC-95,466 NSC-95466 PCNU Urea, 1-(2-chloroethyl)-3-(2,6-dioxo-3-piperidyl)-1-nitroso- 1-(2-Chloroethyl)-3-(2,6-dioxo-3-piperidyl)-1-nitroso- 3-((1-(2-Chloroethyl)-2-oxohydrazino)carbonyl)amino)-2,6-piperidinedione
<b>Inchi:</b>	InChI=1S/C8H11ClN4O4/c9-3-4-13(12-17)8(16)10-5-1-2-6(14)11-7(5)15/h5H,1-4H2,(H,1
<b>InchiKey:</b>	KHWIRCOLWPNBJP-UHFFFAOYSA-N
<b>Formula:</b>	C8H11ClN4O4
<b>SMILES:</b>	O=NN(CCCl)C(=O)NC1CCC(=O)NC1=O
<b>Mol. weight [g/mol]:</b>	262.65
<b>CAS:</b>	13909-02-9

## Physical Properties

Property code	Value	Unit	Source
hf	-567.23	kJ/mol	Joback Method
hvap	77.79	kJ/mol	Joback Method
log10ws	-1.77		Crippen Method
logp	-0.277		Crippen Method
mcvol	171.160	ml/mol	McGowan Method
pc	3598.56	kPa	Joback Method
tb	803.49	K	Joback Method
tc	1038.72	K	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13909029&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13909029&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

# Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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