

# [1,4]Dioxino[2,3-b]-1,4-dioxin, hexahydro-

<b>Other names:</b>	p-Dioxino[2,3-b]-p-dioxin, hexahydro- Ethanedial, cyclic 1,2:1,2-bis(1,2-ethanediyl acetal) Naphthodioxane 1,4,5,8-Naphthodioxane 1,4,5,8-Tetraoxaperhydronaphthalene 1,4,5,8-Tetraoxa-decalin 1,4,5,8-Tetraoxabicyclo[4.4.0]decane Hexahydro[1,4]dioxino[2,3-b][1,4]dioxine
<b>Inchi:</b>	InChI=1S/C6H10O4/c1-2-8-6-5(7-1)9-3-4-10-6/h5-6H,1-4H2
<b>InchiKey:</b>	XIEQDDL BXUOZOD-UHFFFAOYSA-N
<b>Formula:</b>	C6H10O4
<b>SMILES:</b>	C1COC2OCCOC2O1
<b>Mol. weight [g/mol]:</b>	146.14
<b>CAS:</b>	4362-05-4

## Physical Properties

Property code	Value	Unit	Source
gf	-271.74	kJ/mol	Joback Method
hf	-574.21	kJ/mol	Joback Method
hfus	31.08	kJ/mol	Joback Method
hvap	47.50	kJ/mol	Joback Method
ie	9.22	eV	NIST Webbook
ie	8.50	eV	NIST Webbook
log10ws	0.31		Crippen Method
logp	-0.268		Crippen Method
mcvol	97.160	ml/mol	McGowan Method
pc	4627.70	kPa	Joback Method
ripol	1827.00		NIST Webbook
ripol	1827.00		NIST Webbook
tb	475.04	K	Joback Method
tc	705.60	K	Joback Method
tf	285.46	K	Joback Method
vc	0.338	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	235.71	J/molxK	475.04	Joback Method
cpg	251.30	J/molxK	513.47	Joback Method
cpg	265.88	J/molxK	551.89	Joback Method
cpg	279.49	J/molxK	590.32	Joback Method
cpg	292.17	J/molxK	628.75	Joback Method
cpg	303.94	J/molxK	667.17	Joback Method
cpg	314.86	J/molxK	705.60	Joback Method
dvisc	0.0061087	Paxs	285.46	Joback Method
dvisc	0.0034886	Paxs	317.06	Joback Method
dvisc	0.0022053	Paxs	348.65	Joback Method
dvisc	0.0015044	Paxs	380.25	Joback Method
dvisc	0.0010883	Paxs	411.85	Joback Method
dvisc	0.0008245	Paxs	443.44	Joback Method
dvisc	0.0006481	Paxs	475.04	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4362054&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4362054&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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