

# Diethylmalonic acid, butyl 2,4,4-trimethylpentyl ester

Inchi:	InChI=1S/C19H36O4/c1-8-11-12-22-16(20)19(9-2,10-3)17(21)23-14-15(4)13-18(5,6)7/h1
InchiKey:	YXYPVKHNYAXUIS-UHFFFAOYSA-N
Formula:	C19H36O4
SMILES:	CCCCOC(=O)C(CC)(CC)C(=O)OCC(C)CC(C)(C)C
Mol. weight [g/mol]:	328.49

## Physical Properties

Property code	Value	Unit	Source
gf	-355.50	kJ/mol	Joback Method
hf	-947.87	kJ/mol	Joback Method
hfus	32.19	kJ/mol	Joback Method
hvap	73.22	kJ/mol	Joback Method
log10ws	-4.78		Crippen Method
logp	4.752		Crippen Method
mcvol	293.450	ml/mol	McGowan Method
pc	1185.79	kPa	Joback Method
rinsol	1829.00		NIST Webbook
tb	779.80	K	Joback Method
tc	969.61	K	Joback Method
tf	438.05	K	Joback Method
vc	1.119	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	912.00	J/molxK	779.80	Joback Method
cpg	930.22	J/molxK	811.44	Joback Method
cpg	947.36	J/molxK	843.07	Joback Method
cpg	963.48	J/molxK	874.71	Joback Method
cpg	978.62	J/molxK	906.34	Joback Method
cpg	992.81	J/molxK	937.98	Joback Method
cpg	1006.10	J/molxK	969.61	Joback Method
dvisc	0.0010640	Paxs	438.05	Joback Method
dvisc	0.0004408	Paxs	495.01	Joback Method

dvisc	0.0002190	Paxs	551.97	Joback Method
dvisc	0.0001240	Paxs	608.92	Joback Method
dvisc	0.0000774	Paxs	665.88	Joback Method
dvisc	0.0000521	Paxs	722.84	Joback Method
dvisc	0.0000371	Paxs	779.80	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369476&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369476&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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