

# trichloroacetic acid, stearyl ester

<b>Other names:</b>	Octadecyl trichloroacetate
<b>Inchi:</b>	InChI=1S/C20H37Cl3O2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-25-19(24)20(2
<b>InchiKey:</b>	SHVKFBUBFSLWFV-UHFFFAOYSA-N
<b>Formula:</b>	C20H37Cl3O2
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCCOC(=O)C(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	415.87

## Physical Properties

Property code	Value	Unit	Source
gf	-149.35	kJ/mol	Joback Method
hf	-756.90	kJ/mol	Joback Method
hfus	55.52	kJ/mol	Joback Method
hvap	81.13	kJ/mol	Joback Method
log10ws	-8.62		Crippen Method
logp	8.161		Crippen Method
mcvol	336.820	ml/mol	McGowan Method
pc	986.40	kPa	Joback Method
rinpol	2571.00		NIST Webbook
rinpol	2588.00		NIST Webbook
rinpol	2571.00		NIST Webbook
ripol	2987.00		NIST Webbook
tb	842.35	K	Joback Method
tc	1034.98	K	Joback Method
tf	479.50	K	Joback Method
vc	1.315	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1009.89	J/molxK	842.35	Joback Method
cpg	1086.12	J/molxK	1002.88	Joback Method
cpg	1072.65	J/molxK	970.77	Joback Method
cpg	1058.34	J/molxK	938.67	Joback Method
cpg	1043.14	J/molxK	906.56	Joback Method

cpg	1027.01	J/molxK	874.46	Joback Method
cpg	1098.81	J/molxK	1034.98	Joback Method
dvisc	0.0000369	Paxs	842.35	Joback Method
dvisc	0.0000500	Paxs	781.88	Joback Method
dvisc	0.0000714	Paxs	721.40	Joback Method
dvisc	0.0001087	Paxs	660.92	Joback Method
dvisc	0.0001802	Paxs	600.45	Joback Method
dvisc	0.0003345	Paxs	539.98	Joback Method
dvisc	0.0007258	Paxs	479.50	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R146231&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R146231&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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