

# 4-(2,4,6-trimethylphenyl)butan-2-ol

<b>Inchi:</b>	InChI=1S/C13H20O/c1-9-7-10(2)13(11(3)8-9)6-5-12(4)14/h7-8,12,14H,5-6H2,1-4H3
<b>InchiKey:</b>	HAIYXQAXONYOHC-UHFFFAOYSA-N
<b>Formula:</b>	C13H20O
<b>SMILES:</b>	Cc1cc(C)c(CCC(C)O)c(C)c1
<b>Mol. weight [g/mol]:</b>	192.30

## Physical Properties

Property code	Value	Unit	Source
gf	2.84	kJ/mol	Joback Method
hf	-267.04	kJ/mol	Joback Method
hfus	22.86	kJ/mol	Joback Method
hvap	65.08	kJ/mol	Joback Method
log10ws	-3.91		Crippen Method
logp	2.925		Crippen Method
mcvol	176.140	ml/mol	McGowan Method
pc	2311.39	kPa	Joback Method
rinpol	1540.00		NIST Webbook
rinpol	1540.00		NIST Webbook
tb	630.20	K	Joback Method
tc	822.14	K	Joback Method
tf	346.07	K	Joback Method
vc	0.668	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	456.00	J/molxK	630.20	Joback Method
cpg	470.21	J/molxK	662.19	Joback Method
cpg	483.72	J/molxK	694.18	Joback Method
cpg	496.55	J/molxK	726.17	Joback Method
cpg	508.72	J/molxK	758.16	Joback Method
cpg	520.26	J/molxK	790.15	Joback Method
cpg	531.18	J/molxK	822.14	Joback Method
dvisc	0.0034190	Paxs	346.07	Joback Method

dvisc	0.0011346	Paxs	393.43	Joback Method
dvisc	0.0004772	Paxs	440.78	Joback Method
dvisc	0.0002374	Paxs	488.14	Joback Method
dvisc	0.0001337	Paxs	535.49	Joback Method
dvisc	0.0000826	Paxs	582.85	Joback Method
dvisc	0.0000549	Paxs	630.20	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R217328&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R217328&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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