

# 2-Cyclohexen-1-one, 3-methyl-6-(1-methylethenyl)-

Other names:	p-Mentha-1,8-dien-3-one Isopiperitenon Isopiperitenone
Inchi:	InChI=1S/C10H14O/c1-7(2)9-5-4-8(3)6-10(9)11/h6,9H,1,4-5H2,2-3H3
InchiKey:	SEZLYIWMVRUIKT-UHFFFAOYSA-N
Formula:	C10H14O
SMILES:	<chem>C=C(C)C1CCC(C)=CC1=O</chem>
Mol. weight [g/mol]:	150.22
CAS:	529-01-1

## Physical Properties

Property code	Value	Unit	Source
gf	34.80	kJ/mol	Joback Method
hf	-171.16	kJ/mol	Joback Method
hfus	11.24	kJ/mol	Joback Method
hvap	42.89	kJ/mol	Joback Method
ie	9.53	eV	NIST Webbook
log10ws	-2.65		Crippen Method
logp	2.488		Crippen Method
mcvol	133.870	ml/mol	McGowan Method
pc	2859.68	kPa	Joback Method
rinpol	1222.00		NIST Webbook
rinpol	1249.00		NIST Webbook
rinpol	1272.00		NIST Webbook
rinpol	1239.00		NIST Webbook
rinpol	1222.00		NIST Webbook
rinpol	1285.00		NIST Webbook
rinpol	1269.00		NIST Webbook
rinpol	1249.00		NIST Webbook
rinpol	1271.00		NIST Webbook
rinpol	1272.00		NIST Webbook
rinpol	1285.00		NIST Webbook
rinpol	1272.00		NIST Webbook
rinpol	1266.00		NIST Webbook
ripol	1865.00		NIST Webbook
ripol	1851.00		NIST Webbook
ripol	1832.00		NIST Webbook

ripol	1865.00		NIST Webbook
ripol	1865.00		NIST Webbook
ripol	1865.00		NIST Webbook
ripol	1833.00		NIST Webbook
ripol	1865.00		NIST Webbook
tb	516.27	K	Joback Method
tc	742.23	K	Joback Method
tf	275.62	K	Joback Method
vc	0.503	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	302.34	J/mol×K	516.27	Joback Method
cpg	319.31	J/mol×K	553.93	Joback Method
cpg	335.43	J/mol×K	591.59	Joback Method
cpg	350.72	J/mol×K	629.25	Joback Method
cpg	365.16	J/mol×K	666.91	Joback Method
cpg	378.76	J/mol×K	704.57	Joback Method
cpg	391.51	J/mol×K	742.23	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C529011&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C529011&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307i">http://pubs.acs.org/doi/abs/10.1021/ci990307i</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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