

Acetic acid, (2,4-dichlorophenoxy)-, 1-methylethyl ester

Other names:

(2,4-Dichlorophenoxy)acetic acid isopropyl ester
2,4-D isopropyl ester
Acetic acid, (2,4-dichlorophenoxy)-, isopropyl ester
Esteron 44
Isopropyl (2,4-dichlorophenoxy)acetate
Isopropyl 2,4-D ester
Isopropylester kyseliny 2,4-dichlorfenoxyoctove
Weedone 128

Inchi:

InChI=1S/C11H12Cl2O3/c1-7(2)16-11(14)6-15-10-4-3-8(12)5-9(10)13/h3-5,7H,6H2,1-2H

InchiKey:

WHOKDONDRZNCBC-UHFFFAOYSA-N

Formula:

C11H12Cl2O3

SMILES:

CC(C)OC(=O)COc1ccc(Cl)cc1Cl

Mol. weight [g/mol]:

263.12

CAS:

94-11-1

Physical Properties

Property code	Value	Unit	Source
gf	-230.33	kJ/mol	Joback Method
hf	-470.56	kJ/mol	Joback Method
hfus	26.36	kJ/mol	Joback Method
hvap	63.63	kJ/mol	Joback Method
log10ws	-3.61		Crippen Method
logp	3.324		Crippen Method
mcvol	179.880	ml/mol	McGowan Method
pc	2475.19	kPa	Joback Method
rinpol	1700.00		NIST Webbook
rinpol	1700.00		NIST Webbook
tb	660.85	K	Joback Method
tc	881.40	K	Joback Method
tf	404.42	K	Joback Method
vc	0.677	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	421.47	J/molxK	660.85	Joback Method
cpg	433.77	J/molxK	697.61	Joback Method
cpg	445.29	J/molxK	734.37	Joback Method
cpg	456.03	J/molxK	771.12	Joback Method
cpg	465.99	J/molxK	807.88	Joback Method
cpg	475.17	J/molxK	844.64	Joback Method
cpg	483.57	J/molxK	881.40	Joback Method
dvisc	0.0010197	Paxs	404.42	Joback Method
dvisc	0.0006138	Paxs	447.16	Joback Method
dvisc	0.0004037	Paxs	489.90	Joback Method
dvisc	0.0002839	Paxs	532.63	Joback Method
dvisc	0.0002104	Paxs	575.37	Joback Method
dvisc	0.0001626	Paxs	618.11	Joback Method
dvisc	0.0001298	Paxs	660.85	Joback Method
hvapt	69.50	kJ/mol	516.50	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.08352e+01
Coeff. B	-3.00888e+03
Coeff. C	-1.60340e+02
Temperature range (K), min.	445.61
Temperature range (K), max.	705.06

Sources

The Yaws Handbook of Vapor Pressure:
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C94111&Units=SI>

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/55-761-6/Acetic-acid-2-4-dichlorophenoxy-1-methylethyl-ester.pdf>

Generated by Cheméo on 2024-04-23 13:26:31.221987845 +0000 UTC m=+16168040.142565160.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.