

# 2-Oxopentanedioic acid

<b>Other names:</b>	2-Ketoglutaric acid 2-Oxoglutaric acid «alpha»-Ketoglutaric acid Pentanedioic acid, 2-oxo- Glutaric acid, 2-oxo- Glutaric acid, «alpha»-keto- 2-Oxo-1,5-pentanedioic acid NSC 17391 «alpha»-Oxoglutaric acid
<b>Inchi:</b>	InChI=1S/C5H6O5/c6-3(5(9)10)1-2-4(7)8/h1-2H2,(H,7,8)(H,9,10)
<b>InchiKey:</b>	KPGXRSRHYNQIFN-UHFFFAOYSA-N
<b>Formula:</b>	C5H6O5
<b>SMILES:</b>	O=C(O)CCC(=O)C(=O)O
<b>Mol. weight [g/mol]:</b>	146.10
<b>CAS:</b>	328-50-7

## Physical Properties

Property code	Value	Unit	Source
chs	-1801.11	kJ/mol	NIST Webbook
chs	-1798.90 ± 0.84	kJ/mol	NIST Webbook
gf	-669.18	kJ/mol	Joback Method
hf	-788.73	kJ/mol	Joback Method
hfs	-1026.20 ± 0.88	kJ/mol	NIST Webbook
hfs	-1027.89	kJ/mol	NIST Webbook
hfus	21.68	kJ/mol	Joback Method
hvap	80.32	kJ/mol	Joback Method
log10ws	0.61		Crippen Method
logp	-0.495		Crippen Method
mcvol	97.760	ml/mol	McGowan Method
pc	5695.98	kPa	Joback Method
tb	659.77	K	Joback Method
tc	842.08	K	Joback Method
tf	390.90 ± 1.00	K	NIST Webbook
tf	388.45 ± 1.50	K	NIST Webbook
vc	0.371	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	237.97	J/molxK	659.77	Joback Method
cpg	243.35	J/molxK	690.16	Joback Method
cpg	248.42	J/molxK	720.54	Joback Method
cpg	253.19	J/molxK	750.93	Joback Method
cpg	257.66	J/molxK	781.31	Joback Method
cpg	261.84	J/molxK	811.70	Joback Method
cpg	265.74	J/molxK	842.08	Joback Method
dvisc	0.0009585	Paxs	457.91	Joback Method
dvisc	0.0027311	Paxs	417.54	Joback Method
dvisc	0.0003986	Paxs	498.28	Joback Method
dvisc	0.0001891	Paxs	538.65	Joback Method
dvisc	0.0000995	Paxs	579.03	Joback Method
dvisc	0.0000569	Paxs	619.40	Joback Method
dvisc	0.0000349	Paxs	659.77	Joback Method
hfust	28.59	kJ/mol	388.70	NIST Webbook
hsubt	100.00	kJ/mol	277.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C328507&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C328507&amp;Units=SI</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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