

# trans-Phenanthrene, 9,10-dihydro-9,10-diol

<b>Other names:</b>	9,10-Phenanthrenediol, 9,10-dihydro- (E)-
<b>Inchi:</b>	InChI=1S/C14H12O2/c15-13-11-7-3-1-5-9(11)10-6-2-4-8-12(10)14(13)16/h1-8,13-16H
<b>InchiKey:</b>	MFXNBQWUTDDOKE-UHFFFAOYSA-N
<b>Formula:</b>	C14H12O2
<b>SMILES:</b>	OC1c2ccccc2-c2ccccc2C1O
<b>Mol. weight [g/mol]:</b>	212.24
<b>CAS:</b>	572-41-8

## Physical Properties

Property code	Value	Unit	Source
gf	64.06	kJ/mol	Joback Method
hf	-128.01	kJ/mol	Joback Method
hfus	28.80	kJ/mol	Joback Method
hvap	85.42	kJ/mol	Joback Method
ie	8.13	eV	NIST Webbook
log10ws	-4.18		Crippen Method
logp	2.434		Crippen Method
mcvol	161.480	ml/mol	McGowan Method
pc	3568.53	kPa	Joback Method
rinpola	2020.00		NIST Webbook
tb	765.20	K	Joback Method
tc	978.77	K	Joback Method
tf	464.28	K	Joback Method
vc	0.606	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	460.64	J/molxK	765.20	Joback Method
cpg	471.31	J/molxK	800.79	Joback Method
cpg	481.29	J/molxK	836.39	Joback Method
cpg	490.66	J/molxK	871.98	Joback Method
cpg	499.49	J/molxK	907.58	Joback Method
cpg	507.85	J/molxK	943.17	Joback Method

cpg	515.83	J/mol×K	978.77	Joback Method
dvisc	0.0012437	Paxs	464.28	Joback Method
dvisc	0.0005307	Paxs	514.43	Joback Method
dvisc	0.0002635	Paxs	564.59	Joback Method
dvisc	0.0001466	Paxs	614.74	Joback Method
dvisc	0.0000892	Paxs	664.89	Joback Method
dvisc	0.0000581	Paxs	715.05	Joback Method
dvisc	0.0000401	Paxs	765.20	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C572418&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C572418&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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