

# Ethanol, 2-(pentyloxy)-

<b>Other names:</b>	2-(pentyloxy)ethanol 2-hydroxyethyl pentyl ether 2-pentoxyethanol ethanol, 2-pentyloxy- ethylene glycol monopentyl ether pentylglycol
<b>Inchi:</b>	InChI=1S/C7H16O2/c1-2-3-4-6-9-7-5-8/h8H,2-7H2,1H3
<b>InchiKey:</b>	QVQDALFNSIKMBH-UHFFFAOYSA-N
<b>Formula:</b>	C7H16O2
<b>SMILES:</b>	CCCCCOCCO
<b>Mol. weight [g/mol]:</b>	132.20
<b>CAS:</b>	6196-58-3

## Physical Properties

Property code	Value	Unit	Source
gf	-233.76	kJ/mol	Joback Method
hf	-472.26	kJ/mol	Joback Method
hfus	19.16	kJ/mol	Joback Method
hvap	50.27	kJ/mol	Joback Method
log10ws	-1.10		Crippen Method
logp	1.185		Crippen Method
mcvol	121.230	ml/mol	McGowan Method
pc	3035.62	kPa	Joback Method
rinpol	1060.60		NIST Webbook
tb	474.16	K	Joback Method
tc	634.70	K	Joback Method
tf	251.70	K	Joback Method
vc	0.465	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	271.34	J/molxK	474.16	Joback Method
cpg	281.77	J/molxK	500.92	Joback Method

cpg	291.85	J/molxK	527.67	Joback Method
cpg	301.60	J/molxK	554.43	Joback Method
cpg	311.01	J/molxK	581.19	Joback Method
cpg	320.10	J/molxK	607.95	Joback Method
cpg	328.87	J/molxK	634.70	Joback Method
dvisc	0.0300573	Paxs	251.70	Joback Method
dvisc	0.0071642	Paxs	288.78	Joback Method
dvisc	0.0023665	Paxs	325.85	Joback Method
dvisc	0.0009802	Paxs	362.93	Joback Method
dvisc	0.0004781	Paxs	400.01	Joback Method
dvisc	0.0002634	Paxs	437.08	Joback Method
dvisc	0.0001593	Paxs	474.16	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Mutual Solubility and Lower Critical Solution Temperature for Water + Organic Ethanol Systems:</b>	<a href="https://www.doi.org/10.1021/je049635u">https://www.doi.org/10.1021/je049635u</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6196583&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6196583&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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