

# 2,4,6-Trihydroxytoluene

<b>Other names:</b>	1,3,5-Benzenetriol, 2-methyl-Methylphloroglucinol Phloroglucinol, methyl-Toluene-2,4,6-triol 2-Methylphloroglucinol 2-Methyl-1,3,5-benzenetriol
<b>Inchi:</b>	InChI=1S/C7H8O3/c1-4-6(9)2-5(8)3-7(4)10/h2-3,8-10H,1H3
<b>InchiKey:</b>	BPHYZRNTQNPLFI-UHFFFAOYSA-N
<b>Formula:</b>	C7H8O3
<b>SMILES:</b>	<chem>Cc1c(O)cc(O)cc1O</chem>
<b>Mol. weight [g/mol]:</b>	140.14
<b>CAS:</b>	88-03-9

## Physical Properties

Property code	Value	Unit	Source
gf	-343.39	kJ/mol	Joback Method
hf	-483.21	kJ/mol	Joback Method
hfus	25.28	kJ/mol	Joback Method
hvap	72.49	kJ/mol	Joback Method
log10ws	-0.59		Crippen Method
logp	1.112		Crippen Method
mcvol	103.340	ml/mol	McGowan Method
pc	8087.06	kPa	Joback Method
tb	628.10	K	Joback Method
tc	881.41	K	Joback Method
tf	530.23	K	Joback Method
vc	0.217	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	264.27	J/mol×K	628.10	Joback Method
cpg	298.79	J/mol×K	839.19	Joback Method
cpg	292.12	J/mol×K	796.97	Joback Method

cpg	285.55	J/molxK	754.76	Joback Method
cpg	278.87	J/molxK	712.54	Joback Method
cpg	271.85	J/molxK	670.32	Joback Method
cpg	305.80	J/molxK	881.41	Joback Method
dvisc	0.0000008	Paxs	628.10	Joback Method
dvisc	0.0000011	Paxs	611.79	Joback Method
dvisc	0.0000016	Paxs	595.48	Joback Method
dvisc	0.0000024	Paxs	579.16	Joback Method
dvisc	0.0000037	Paxs	562.85	Joback Method
dvisc	0.0000059	Paxs	546.54	Joback Method
dvisc	0.0000095	Paxs	530.23	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C88039&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C88039&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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