

# Isophthalic acid, isohexyl pentachlorophenyl ester

Inchi:	InChI=1S/C20H17Cl5O4/c1-10(2)5-4-8-28-19(26)11-6-3-7-12(9-11)20(27)29-18-16(24)1
InchiKey:	VNSVEQGJUAHDPB-UHFFFAOYSA-N
Formula:	C20H17Cl5O4
SMILES:	CC(C)CCCOC(=O)c1cccc(C(=O)Oc2c(Cl)c(Cl)c(Cl)c(Cl)c2Cl)c1
Mol. weight [g/mol]:	498.61

## Physical Properties

Property code	Value	Unit	Source
gf	-245.37	kJ/mol	Joback Method
hf	-625.47	kJ/mol	Joback Method
hfus	56.34	kJ/mol	Joback Method
hvap	108.49	kJ/mol	Joback Method
log10ws	-9.08		Crippen Method
logp	7.766		Crippen Method
mvol	321.220	ml/mol	McGowan Method
pc	1422.92	kPa	Joback Method
rinpol	3380.00		NIST Webbook
rinpol	3380.00		NIST Webbook
tb	1079.53	K	Joback Method
tc	1330.07	K	Joback Method
tf	722.04	K	Joback Method
vc	1.226	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	869.29	J/mol×K	1079.53	Joback Method
cpg	887.49	J/mol×K	1288.31	Joback Method
cpg	886.68	J/mol×K	1246.55	Joback Method
cpg	884.48	J/mol×K	1204.80	Joback Method
cpg	880.87	J/mol×K	1163.04	Joback Method
cpg	875.81	J/mol×K	1121.29	Joback Method
cpg	886.91	J/mol×K	1330.07	Joback Method
dvisc	0.0000256	Paxs	1079.53	Joback Method

dvisc	0.0000311	Paxs	1019.95	Joback Method
dvisc	0.0000387	Paxs	960.37	Joback Method
dvisc	0.0000495	Paxs	900.78	Joback Method
dvisc	0.0000656	Paxs	841.20	Joback Method
dvisc	0.0000908	Paxs	781.62	Joback Method
dvisc	0.0001325	Paxs	722.04	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U356578&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U356578&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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