

1-Heptadecene, 2,4,6,8,10,12,14-heptamethyl

Inchi:	InChI=1S/C24H48/c1-10-11-19(4)13-21(6)15-23(8)17-24(9)16-22(7)14-20(5)12-18(2)3/h
InchiKey:	XOMQDLODORPXCA-UHFFFAOYSA-N
Formula:	C24H48
SMILES:	C=C(C)CC(C)CC(C)CC(C)CC(C)CC(C)CC(C)CCC
Mol. weight [g/mol]:	336.64

Physical Properties

Property code	Value	Unit	Source
gf	215.85	kJ/mol	Joback Method
hf	-454.73	kJ/mol	Joback Method
hfus	34.19	kJ/mol	Joback Method
hvap	66.10	kJ/mol	Joback Method
log10ws	-8.27		Crippen Method
logp	8.520		Crippen Method
mvol	344.720	ml/mol	McGowan Method
pc	851.47	kPa	Joback Method
rinpol	1973.00		NIST Webbook
rinpol	1973.00		NIST Webbook
tb	742.44	K	Joback Method
tc	919.57	K	Joback Method
tf	254.52	K	Joback Method
vc	1.325	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1063.26	J/mol×K	742.44	Joback Method
cpg	1086.61	J/mol×K	771.96	Joback Method
cpg	1108.81	J/mol×K	801.48	Joback Method
cpg	1129.93	J/mol×K	831.00	Joback Method
cpg	1149.99	J/mol×K	860.53	Joback Method
cpg	1169.05	J/mol×K	890.05	Joback Method
cpg	1187.15	J/mol×K	919.57	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R568131&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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