

# Benzenesulfonic acid, 4-methyl-, butyl ester

<b>Other names:</b>	p-Toluenesulfonic acid, butyl ester Butyl p-methylbenzenesulfonate Butyl p-toluenesulfonate Butyl tosylate Butyl para-toluenesulfonate n-Butyl p-toluenesulfonate Butyl ester of 4-toluenesulfonic acid Butyl 4-methylbenzenesulfonate NSC 6190 butyl toluene-4-sulphonate
<b>Inchi:</b>	InChI=1S/C11H16O3S/c1-3-4-9-14-15(12,13)11-7-5-10(2)6-8-11/h5-8H,3-4,9H2,1-2H3
<b>InchiKey:</b>	QYJXDIUNDMRLAO-UHFFFAOYSA-N
<b>Formula:</b>	C11H16O3S
<b>SMILES:</b>	CCCCOS(=O)(=O)c1ccc(C)cc1
<b>Mol. weight [g/mol]:</b>	228.31
<b>CAS:</b>	778-28-9

## Physical Properties

Property code	Value	Unit	Source
gf	-429.02	kJ/mol	Joback Method
hf	-630.88	kJ/mol	Joback Method
hfus	30.46	kJ/mol	Joback Method
hvap	64.06	kJ/mol	Joback Method
log10ws	-2.97		Crippen Method
logp	2.500		Crippen Method
mcvol	176.050	ml/mol	McGowan Method
pc	3018.96	kPa	Joback Method
tb	552.94	K	Joback Method
tc	748.21	K	Joback Method
tf	313.46	K	Joback Method
vc	0.688	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	408.73	J/mol×K	552.94	Joback Method
cpg	424.05	J/mol×K	585.48	Joback Method
cpg	438.61	J/mol×K	618.03	Joback Method
cpg	452.42	J/mol×K	650.57	Joback Method
cpg	465.46	J/mol×K	683.12	Joback Method
cpg	477.75	J/mol×K	715.66	Joback Method
cpg	489.29	J/mol×K	748.21	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C778289&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C778289&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/56-109-9/Benzenesulfonic-acid-4-methyl-butyl-ester.pdf>

Generated by Cheméo on 2025-12-25 01:09:05.555861753 +0000 UTC m=+6373143.085902407.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.