

# Anisyl heptanoate

<b>Inchi:</b>	InChI=1S/C15H22O3/c1-3-4-5-6-7-15(16)18-12-13-8-10-14(17-2)11-9-13/h8-11H,3-7,12H
<b>InchiKey:</b>	WXFXWKOGMWSKOV-UHFFFAOYSA-N
<b>Formula:</b>	C15H22O3
<b>SMILES:</b>	CCCCCCC(=O)OCc1ccc(OC)cc1
<b>Mol. weight [g/mol]:</b>	250.33

## Physical Properties

Property code	Value	Unit	Source
gf	-160.72	kJ/mol	Joback Method
hf	-504.89	kJ/mol	Joback Method
hfus	32.23	kJ/mol	Joback Method
hvap	63.49	kJ/mol	Joback Method
log10ws	-4.26		Crippen Method
logp	3.709		Crippen Method
mcvol	211.760	ml/mol	McGowan Method
pc	1869.17	kPa	Joback Method
rinpol	1862.00		NIST Webbook
rinpol	1862.00		NIST Webbook
ripol	2600.00		NIST Webbook
tb	672.97	K	Joback Method
tc	869.94	K	Joback Method
tf	392.14	K	Joback Method
vc	0.809	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	574.00	J/molxK	672.97	Joback Method
cpg	590.26	J/molxK	705.80	Joback Method
cpg	605.64	J/molxK	738.63	Joback Method
cpg	620.13	J/molxK	771.45	Joback Method
cpg	633.77	J/molxK	804.28	Joback Method
cpg	646.54	J/molxK	837.11	Joback Method
cpg	658.47	J/molxK	869.94	Joback Method

dvisc	0.0010931	Paxs	392.14	Joback Method
dvisc	0.0006017	Paxs	438.95	Joback Method
dvisc	0.0003716	Paxs	485.75	Joback Method
dvisc	0.0002498	Paxs	532.56	Joback Method
dvisc	0.0001790	Paxs	579.36	Joback Method
dvisc	0.0001349	Paxs	626.17	Joback Method
dvisc	0.0001057	Paxs	672.97	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R409433&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R409433&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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