

# Cholestane

**Other names:**

Cholestane, (5«alpha»)-  
5«alpha»-Cholestane  
«alpha»-Cholestane  
28,29,30-Trinorlanostane  
10,13-Dimethyl-17-(1',5'-dimethylhexyl)-hexadecahydrocyclopenta(a)phenanthrene  
5A-Cholestane, TMS

**Inchi:** C[C@H](CCCC(C)C)[C@H]1CC[C@@H]2[C@@]1(CC[C@H]3[C@H]2CC=C4[C@@]3(CC[C@@H](C4)C)C)C**InchiKey:** XIIAYQZJNBULGD-BTSBFKJRSA-N**Formula:** C<sub>27</sub>H<sub>48</sub>**SMILES:** CC(C)CCCC(C)C1CCC2C3CCC4CCCCC4(C)C3CCC12C**Mol. weight [g/mol]:** 372.67**CAS:** 481-21-0

## Physical Properties

Property code	Value	Unit	Source
gf	319.97	kJ/mol	Joback Method
hf	-381.31	kJ/mol	Joback Method
hfus	31.30	kJ/mol	Joback Method
hsub	133.80	kJ/mol	NIST Webbook
hvap	72.20	kJ/mol	Joback Method
log10ws	-8.53		Crippen Method
logp	8.498		Crippen Method
mcvol	347.850	ml/mol	McGowan Method
pc	1002.71	kPa	Joback Method
rinpol	2852.00		NIST Webbook
rinpol	2819.00		NIST Webbook
rinpol	2838.00		NIST Webbook
rinpol	2856.00		NIST Webbook
rinpol	2852.00		NIST Webbook
rinpol	2866.00		NIST Webbook
rinpol	2856.00		NIST Webbook
rinpol	2852.00		NIST Webbook
rinpol	2857.00		NIST Webbook
rinpol	2857.00		NIST Webbook
rinpol	2866.00		NIST Webbook
tb	851.06	K	Joback Method

tc	1072.72	K	Joback Method
tf	453.29	K	Joback Method
vc	1.317	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1250.75	J/mol×K	851.06	Joback Method
cpg	1282.08	J/mol×K	888.00	Joback Method
cpg	1312.99	J/mol×K	924.95	Joback Method
cpg	1343.81	J/mol×K	961.89	Joback Method
cpg	1374.86	J/mol×K	998.83	Joback Method
cpg	1406.45	J/mol×K	1035.78	Joback Method
cpg	1438.92	J/mol×K	1072.72	Joback Method
hfust	25.40	kJ/mol	351.80	NIST Webbook
hvapt	108.40	kJ/mol	352.00	NIST Webbook
hvapt	115.60	kJ/mol	509.50	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C481210&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C481210&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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