

# 1,3-Dioxolane, 2-ethyl-

<b>Other names:</b>	Propanal, cyclic 1,2-ethanediyl acetal 2-Ethyl-1,3-dioxolane
<b>Inchi:</b>	InChI=1S/C5H10O2/c1-2-5-6-3-4-7-5/h5H,2-4H2,1H3
<b>InchiKey:</b>	WEBRDDDJKLAXTO-UHFFFAOYSA-N
<b>Formula:</b>	C5H10O2
<b>SMILES:</b>	CCC1OCCO1
<b>Mol. weight [g/mol]:</b>	102.13
<b>CAS:</b>	2568-96-9

## Physical Properties

Property code	Value	Unit	Source
gf	-144.47	kJ/mol	Joback Method
hf	-350.05	kJ/mol	Joback Method
hfus	18.60	kJ/mol	Joback Method
hvap	36.00	kJ/mol	Joback Method
log10ws	-0.59		Crippen Method
logp	0.769		Crippen Method
mvol	82.190	ml/mol	McGowan Method
pc	4238.55	kPa	Joback Method
rinpol	774.00		NIST Webbook
rinpol	730.00		NIST Webbook
rinpol	730.00		NIST Webbook
rinpol	738.00		NIST Webbook
rinpol	735.00		NIST Webbook
rinpol	787.00		NIST Webbook
tb	379.00 ± 2.00	K	NIST Webbook
tc	581.47	K	Joback Method
tf	210.15	K	Joback Method
vc	0.298	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	156.96	J/mol×K	382.98	Joback Method

cpg	211.03	J/mol×K	548.39	Joback Method
cpg	201.29	J/mol×K	515.31	Joback Method
cpg	191.03	J/mol×K	482.23	Joback Method
cpg	180.23	J/mol×K	449.14	Joback Method
cpg	168.88	J/mol×K	416.06	Joback Method
cpg	220.26	J/mol×K	581.47	Joback Method
dvisc	0.0004296	Paxs	382.98	Joback Method
dvisc	0.0005479	Paxs	354.17	Joback Method
dvisc	0.0007296	Paxs	325.37	Joback Method
dvisc	0.0010272	Paxs	296.56	Joback Method
dvisc	0.0015566	Paxs	267.76	Joback Method
dvisc	0.0026075	Paxs	238.95	Joback Method
dvisc	0.0050314	Paxs	210.15	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2568969&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2568969&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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