

# Naphthalene, 1,7-dimethyl-

<b>Other names:</b>	1,7-Dimethylnaphthalene Naphthalene, 3,5-dimethyl
<b>Inchi:</b>	InChI=1S/C12H12/c1-9-6-7-11-5-3-4-10(2)12(11)8-9/h3-8H,1-2H3
<b>InchiKey:</b>	SPUWFVKLHHEKGV-UHFFFAOYSA-N
<b>Formula:</b>	C12H12
<b>SMILES:</b>	<chem>Cc1ccc2cccc(C)c2c1</chem>
<b>Mol. weight [g/mol]:</b>	156.22
<b>CAS:</b>	575-37-1

## Physical Properties

Property code	Value	Unit	Source
af	0.4430		KDB
gf	249.96	kJ/mol	Joback Method
hf	113.65	kJ/mol	Joback Method
hfus	17.12	kJ/mol	Joback Method
hvap	47.55	kJ/mol	Joback Method
log10ws	-4.23		Crippen Method
logp	3.457		Crippen Method
mcvol	136.720	ml/mol	McGowan Method
pc	3010.00	kPa	KDB
rinpol	243.20		NIST Webbook
rinpol	1396.00		NIST Webbook
rinpol	1418.70		NIST Webbook
rinpol	1409.00		NIST Webbook
rinpol	1414.00		NIST Webbook
rinpol	1418.00		NIST Webbook
rinpol	1386.00		NIST Webbook
rinpol	1409.50		NIST Webbook
rinpol	1410.00		NIST Webbook
rinpol	242.99		NIST Webbook
rinpol	240.66		NIST Webbook
rinpol	1410.00		NIST Webbook
rinpol	243.75		NIST Webbook
rinpol	243.05		NIST Webbook
rinpol	242.19		NIST Webbook
rinpol	242.90		NIST Webbook
rinpol	242.77		NIST Webbook

rinpol	240.66		NIST Webbook
rinpol	243.80		NIST Webbook
rinpol	1422.40		NIST Webbook
rinpol	244.29		NIST Webbook
rinpol	242.60		NIST Webbook
rinpol	244.10		NIST Webbook
rinpol	242.80		NIST Webbook
rinpol	242.80		NIST Webbook
rinpol	1396.00		NIST Webbook
rinpol	1409.00		NIST Webbook
rinpol	1415.00		NIST Webbook
rinpol	1414.00		NIST Webbook
rinpol	1414.00		NIST Webbook
rinpol	1389.60		NIST Webbook
rinpol	1423.00		NIST Webbook
rinpol	240.70		NIST Webbook
rinpol	1396.00		NIST Webbook
ripol	2000.00		NIST Webbook
ripol	2000.00		NIST Webbook
ripol	2000.00		NIST Webbook
tb	531.00 ± 4.00	K	NIST Webbook
tb	536.20	K	KDB
tb	535.00 ± 4.00	K	NIST Webbook
tb	536.00 ± 4.00	K	NIST Webbook
tb	535.90 ± 2.00	K	NIST Webbook
tb	536.00 ± 1.00	K	NIST Webbook
tc	770.60	K	KDB
tf	260.00	K	KDB
vc	0.521	m <sup>3</sup> /kmol	KDB
zc	0.2449940		KDB

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	296.31	J/mol×K	529.58	Joback Method
cpg	361.06	J/mol×K	722.55	Joback Method
cpg	349.89	J/mol×K	683.95	Joback Method
cpg	337.89	J/mol×K	645.36	Joback Method
cpg	325.00	J/mol×K	606.77	Joback Method
cpg	311.16	J/mol×K	568.17	Joback Method
cpg	371.48	J/mol×K	761.14	Joback Method

dvisc	0.0002960	Paxs	529.58	Joback Method
dvisc	0.0003443	Paxs	492.84	Joback Method
dvisc	0.0004103	Paxs	456.11	Joback Method
dvisc	0.0005043	Paxs	419.37	Joback Method
dvisc	0.0006448	Paxs	382.63	Joback Method
dvisc	0.0008686	Paxs	345.90	Joback Method
dvisc	0.0012561	Paxs	309.16	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	421.20	K	2.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.40634e+01
Coeff. B	-3.94601e+03
Coeff. C	-1.18215e+02
Temperature range (K), min.	404.66
Temperature range (K), max.	569.09

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol775.mol">https://www.thermo.com/files/research/kdb/mol/mol775.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C575371&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C575371&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

# Legend

<b>af:</b>	Acentric Factor
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume
<b>zc:</b>	Critical Compressibility

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