

# 7-Tetradecyne

<b>Other names:</b>	7-C14H26 7-Tetadecyne
<b>Inchi:</b>	InChI=1S/C14H26/c1-3-5-7-9-11-13-14-12-10-8-6-4-2/h3-12H2,1-2H3
<b>InchiKey:</b>	AFNWSIIBAYUTTTL-UHFFFAOYSA-N
<b>Formula:</b>	C14H26
<b>SMILES:</b>	CCCCCCC#CCCCCCC
<b>Mol. weight [g/mol]:</b>	194.36
<b>CAS:</b>	35216-11-6

## Physical Properties

Property code	Value	Unit	Source
gf	269.80	kJ/mol	Joback Method
hf	-59.99	kJ/mol	Joback Method
hfus	35.14	kJ/mol	Joback Method
hvap	48.91	kJ/mol	Joback Method
ie	9.03 ± 0.04	eV	NIST Webbook
ie	9.07 ± 0.01	eV	NIST Webbook
log10ws	-5.48		Crippen Method
logp	4.931		Crippen Method
mcvol	199.520	ml/mol	McGowan Method
pc	1733.22	kPa	Joback Method
ripol	1400.00		NIST Webbook
ripol	1400.00		NIST Webbook
ripol	1401.00		NIST Webbook
ripol	1416.00		NIST Webbook
ripol	1416.00		NIST Webbook
ripol	1562.30		NIST Webbook
ripol	1562.30		NIST Webbook
ripol	1545.10		NIST Webbook
ripol	1574.00		NIST Webbook
ripol	1545.10		NIST Webbook
ripol	1562.30		NIST Webbook
ripol	1563.00		NIST Webbook
ripol	1572.00		NIST Webbook
ripol	1572.00		NIST Webbook
ripol	1574.00		NIST Webbook
ripol	1574.00		NIST Webbook

ripol	1575.00		NIST Webbook
ripol	1541.30		NIST Webbook
ripol	1545.10		NIST Webbook
tb	528.72	K	Joback Method
tc	706.35	K	Joback Method
tf	353.64	K	Joback Method
vc	0.781	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	473.12	J/mol×K	528.72	Joback Method
cpg	490.92	J/mol×K	558.32	Joback Method
cpg	507.98	J/mol×K	587.93	Joback Method
cpg	524.33	J/mol×K	617.53	Joback Method
cpg	539.98	J/mol×K	647.14	Joback Method
cpg	554.96	J/mol×K	676.74	Joback Method
cpg	569.28	J/mol×K	706.35	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	397.00	K	2.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.65575e+01
Coeff. B	-5.18082e+03
Coeff. C	-9.00450e+01
Temperature range (K), min.	408.48
Temperature range (K), max.	550.73

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C35216116&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C35216116&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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