

# Sebacic acid, pentyl 2-phenylphenyl ester

<b>Inchi:</b>	InChI=1S/C27H36O4/c1-2-3-15-22-30-26(28)20-11-6-4-5-7-12-21-27(29)31-25-19-14-13
<b>InchiKey:</b>	AHLVYHNNJAEULO-UHFFFAOYSA-N
<b>Formula:</b>	C27H36O4
<b>SMILES:</b>	CCCCCOC(=O)CCCCCCCC(=O)Oc1ccccc1-c1ccccc1
<b>Mol. weight [g/mol]:</b>	424.57

## Physical Properties

Property code	Value	Unit	Source
gf	-76.19	kJ/mol	Joback Method
hf	-628.62	kJ/mol	Joback Method
hfus	58.95	kJ/mol	Joback Method
hvap	99.22	kJ/mol	Joback Method
log10ws	-8.69		Crippen Method
logp	7.113		Crippen Method
mvol	358.650	ml/mol	McGowan Method
pc	1060.33	kPa	Joback Method
rinpol	3174.00		NIST Webbook
rinpol	3174.00		NIST Webbook
tb	1028.08	K	Joback Method
tc	1259.02	K	Joback Method
tf	603.73	K	Joback Method
vc	1.379	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1198.95	J/molxK	1028.08	Joback Method
cpg	1213.64	J/molxK	1066.57	Joback Method
cpg	1226.81	J/molxK	1105.06	Joback Method
cpg	1238.53	J/molxK	1143.55	Joback Method
cpg	1248.86	J/molxK	1182.04	Joback Method
cpg	1257.89	J/molxK	1220.53	Joback Method
cpg	1265.69	J/molxK	1259.02	Joback Method
dvisc	0.0002386	Paxs	603.73	Joback Method

dvisc	0.0001276	Paxs	674.46	Joback Method
dvisc	0.0000768	Paxs	745.18	Joback Method
dvisc	0.0000505	Paxs	815.90	Joback Method
dvisc	0.0000355	Paxs	886.63	Joback Method
dvisc	0.0000263	Paxs	957.35	Joback Method
dvisc	0.0000203	Paxs	1028.08	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U355065&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U355065&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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