

# Acetic acid, (acetyloxy)-

<b>Other names:</b>	Glycolic acid, acetate (Acetyloxy)acetic acid «alpha»-Acetoxyacetic acid 2-Hydroxyacetic acid acetata Acetoxyacetic acid Acetylglycolic acid O-Acetylglycolic acid Acetic acid, 2-(acetyloxy)-
<b>Inchi:</b>	InChI=1S/C4H6O4/c1-3(5)8-2-4(6)7/h2H2,1H3,(H,6,7)
<b>InchiKey:</b>	MLXDUYUQINCFV-UHFFFAOYSA-N
<b>Formula:</b>	C4H6O4
<b>SMILES:</b>	CC(=O)OCC(=O)O
<b>Mol. weight [g/mol]:</b>	118.09
<b>CAS:</b>	13831-30-6

## Physical Properties

Property code	Value	Unit	Source
gf	-516.86	kJ/mol	Joback Method
hf	-635.50	kJ/mol	Joback Method
hfus	14.59	kJ/mol	Joback Method
hvap	57.08	kJ/mol	Joback Method
log10ws	0.54		Crippen Method
logp	-0.366		Crippen Method
mcvol	82.100	ml/mol	McGowan Method
pc	5094.76	kPa	Joback Method
rinpol	1019.80		NIST Webbook
tb	513.26	K	Joback Method
tc	696.47	K	Joback Method
tf	340.00 ± 1.00	K	NIST Webbook
vc	0.308	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	174.45	J/molxK	513.26	Joback Method
cpg	180.44	J/molxK	543.80	Joback Method
cpg	186.21	J/molxK	574.33	Joback Method
cpg	191.75	J/molxK	604.87	Joback Method
cpg	197.06	J/molxK	635.40	Joback Method
cpg	202.13	J/molxK	665.94	Joback Method
cpg	206.96	J/molxK	696.47	Joback Method
dvisc	0.0068490	Paxs	317.75	Joback Method
dvisc	0.0028285	Paxs	350.33	Joback Method
dvisc	0.0013579	Paxs	382.92	Joback Method
dvisc	0.0007314	Paxs	415.50	Joback Method
dvisc	0.0004310	Paxs	448.09	Joback Method
dvisc	0.0002729	Paxs	480.68	Joback Method
dvisc	0.0001831	Paxs	513.26	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	414.70	K	1.60	NIST Webbook
tbrp	417.70	K	1.60	NIST Webbook
tbrp	418.00	K	1.60	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13831306&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13831306&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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