

# Ethanal, 2-(4-hydroxy-2,6-dimethoxyphenyl)

<b>Inchi:</b>	InChI=1S/C9H10O4/c1-12-8-3-6(11)4-9(13-2)7(8)5-10/h3-5,11H,1-2H3
<b>InchiKey:</b>	HZWPJAZIRZFCGX-UHFFFAOYSA-N
<b>Formula:</b>	C9H10O4
<b>SMILES:</b>	<chem>COc1cc(O)cc(OC)c1C=O</chem>
<b>Mol. weight [g/mol]:</b>	182.17

## Physical Properties

Property code	Value	Unit	Source
gf	-346.09	kJ/mol	Joback Method
hf	-542.83	kJ/mol	Joback Method
hfus	22.78	kJ/mol	Joback Method
hvap	63.78	kJ/mol	Joback Method
log10ws	-1.50		Crippen Method
logp	1.222		Crippen Method
mcvol	133.090	ml/mol	McGowan Method
pc	3945.61	kPa	Joback Method
rinpol	1642.00		NIST Webbook
rinpol	1642.00		NIST Webbook
tb	616.08	K	Joback Method
tc	837.67	K	Joback Method
tf	440.83	K	Joback Method
vc	0.451	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	325.35	J/molxK	616.08	Joback Method
cpg	335.72	J/molxK	653.01	Joback Method
cpg	345.52	J/molxK	689.94	Joback Method
cpg	354.78	J/molxK	726.88	Joback Method
cpg	363.52	J/molxK	763.81	Joback Method
cpg	371.79	J/molxK	800.74	Joback Method
cpg	379.60	J/molxK	837.67	Joback Method
dvisc	0.0003979	Paxs	440.83	Joback Method

dvisc	0.0002252	Paxs	470.04	Joback Method
dvisc	0.0001362	Paxs	499.25	Joback Method
dvisc	0.0000871	Paxs	528.45	Joback Method
dvisc	0.0000584	Paxs	557.66	Joback Method
dvisc	0.0000407	Paxs	586.87	Joback Method
dvisc	0.0000294	Paxs	616.08	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R88457&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R88457&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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