

# 2-[2-(2-Butoxyethoxy)ethoxy]ethyl 2,2,3,3,4,4,4-heptafluorobutanoate

**Other names:** Triethylene glycol monobutyl ether, heptafluorobutyrate

3,6,9-Trioxatridec-1-yl heptafluorobutyrate

**Inchi:** InChI=1S/C14H21F7O5/c1-2-3-4-23-5-6-24-7-8-25-9-10-26-11(22)12(15,16)13(17,18)14

**InchiKey:** BZTFUNZLKSUPGB-UHFFFAOYSA-N

**Formula:** C14H21F7O5

**SMILES:** CCCCOCOCOCOC(=O)C(F)(F)C(F)(F)C(F)(F)F

**Mol. weight [g/mol]:** 402.30

## Physical Properties

Property code	Value	Unit	Source
gf	-1837.07	kJ/mol	Joback Method
hf	-2372.77	kJ/mol	Joback Method
hfus	37.68	kJ/mol	Joback Method
hvap	53.54	kJ/mol	Joback Method
log10ws	-3.10		Crippen Method
logp	3.212		Crippen Method
mcvol	245.560	ml/mol	McGowan Method
pc	1253.03	kPa	Joback Method
rinpol	1488.80		NIST Webbook
tb	648.47	K	Joback Method
tc	802.84	K	Joback Method
tf	397.78	K	Joback Method
vc	0.991	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	718.84	J/molxK	648.47	Joback Method
cpg	733.11	J/molxK	674.20	Joback Method
cpg	746.66	J/molxK	699.93	Joback Method
cpg	759.53	J/molxK	725.65	Joback Method
cpg	771.72	J/molxK	751.38	Joback Method
cpg	783.25	J/molxK	777.11	Joback Method
cpg	794.15	J/molxK	802.84	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U352020&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U352020&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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