

4-Methyl-4-heptanolide

Inchi:	InChI=1S/C8H14O2/c1-3-5-8(2)6-4-7(9)10-8/h3-6H2,1-2H3
InchiKey:	YSENGVCQTXRIED-UHFFFAOYSA-N
Formula:	C8H14O2
SMILES:	CCCC1(C)CCC(=O)O1
Mol. weight [g/mol]:	142.20

Physical Properties

Property code	Value	Unit	Source
gf	-161.17	kJ/mol	Joback Method
hf	-402.43	kJ/mol	Joback Method
hfus	11.60	kJ/mol	Joback Method
hvap	41.27	kJ/mol	Joback Method
log10ws	-2.04		Crippen Method
logp	1.882		Crippen Method
mvol	120.160	ml/mol	McGowan Method
pc	3322.01	kPa	Joback Method
ripol	1150.00		NIST Webbook
ripol	1760.00		NIST Webbook
tb	492.73	K	Joback Method
tc	712.64	K	Joback Method
tf	309.51	K	Joback Method
vc	0.451	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	277.93	J/mol×K	492.73	Joback Method
cpg	293.32	J/mol×K	529.38	Joback Method
cpg	307.79	J/mol×K	566.03	Joback Method
cpg	321.45	J/mol×K	602.68	Joback Method
cpg	334.37	J/mol×K	639.33	Joback Method
cpg	346.66	J/mol×K	675.98	Joback Method
cpg	358.40	J/mol×K	712.64	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R434721&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpolar:	Non-polar retention indices
ripolar:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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