

3-ethyl-1,2-dithiolan-4-one

Other names:	1,2-Dithiolan-4-one, 3-ethyl
Inchi:	InChI=1S/C5H8OS2/c1-2-5-4(6)3-7-8-5/h5H,2-3H2,1H3
InchiKey:	TUMVOBRANVUYMK-UHFFFAOYSA-N
Formula:	C5H8OS2
SMILES:	CCC1SSCC1=O
Mol. weight [g/mol]:	148.25

Physical Properties

Property code	Value	Unit	Source
gf	-15.10	kJ/mol	Joback Method
hf	-133.23	kJ/mol	Joback Method
hfus	9.46	kJ/mol	Joback Method
hvap	42.85	kJ/mol	Joback Method
log10ws	-1.96		Crippen Method
logp	1.729		Crippen Method
mcvol	104.720	ml/mol	McGowan Method
pc	4474.22	kPa	Joback Method
rinpol	1169.00		NIST Webbook
rinpol	1167.00		NIST Webbook
rinpol	1163.00		NIST Webbook
tb	492.56	K	Joback Method
tc	745.51	K	Joback Method
tf	392.13	K	Joback Method
vc	0.355	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	208.04	J/mol×K	492.56	Joback Method
cpg	220.32	J/mol×K	534.72	Joback Method
cpg	231.96	J/mol×K	576.88	Joback Method
cpg	242.96	J/mol×K	619.03	Joback Method
cpg	253.30	J/mol×K	661.19	Joback Method
cpg	262.99	J/mol×K	703.35	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R90363&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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