

# Dimethylmalonic acid, isobutyl neopentyl ester

Inchi:	InChI=1S/C14H26O4/c1-10(2)8-17-11(15)14(6,7)12(16)18-9-13(3,4)5/h10H,8-9H2,1-7H3
InchiKey:	LFRIQNIYNUVMTE-UHFFFAOYSA-N
Formula:	C14H26O4
SMILES:	CC(C)COC(=O)C(C)(C)C(=O)OCC(C)(C)C
Mol. weight [g/mol]:	258.35

## Physical Properties

Property code	Value	Unit	Source
gf	-397.60	kJ/mol	Joback Method
hf	-844.67	kJ/mol	Joback Method
hfus	19.24	kJ/mol	Joback Method
hvap	62.09	kJ/mol	Joback Method
log10ws	-2.68		Crippen Method
logp	2.801		Crippen Method
mcvol	223.000	ml/mol	McGowan Method
pc	1701.90	kPa	Joback Method
rinpol	1416.00		NIST Webbook
rinpol	1416.00		NIST Webbook
tb	665.40	K	Joback Method
tc	859.55	K	Joback Method
tf	381.70	K	Joback Method
vc	0.840	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	627.85	J/molxK	665.40	Joback Method
cpg	644.76	J/molxK	697.76	Joback Method
cpg	660.69	J/molxK	730.12	Joback Method
cpg	675.68	J/molxK	762.48	Joback Method
cpg	689.75	J/molxK	794.84	Joback Method
cpg	702.96	J/molxK	827.19	Joback Method
cpg	715.32	J/molxK	859.55	Joback Method
dvisc	0.0020424	Paxs	381.70	Joback Method

dvisc	0.0008881	Paxs	428.98	Joback Method
dvisc	0.0004556	Paxs	476.27	Joback Method
dvisc	0.0002637	Paxs	523.55	Joback Method
dvisc	0.0001671	Paxs	570.83	Joback Method
dvisc	0.0001135	Paxs	618.12	Joback Method
dvisc	0.0000815	Paxs	665.40	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U361744&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U361744&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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