

# 3-Pentenoic acid

<b>Inchi:</b>	InChI=1S/C5H8O2/c1-2-3-4-5(6)7/h2-3H,4H2,1H3,(H,6,7)
<b>InchiKey:</b>	UIUWNILCHFBLEQ-UHFFFAOYSA-N
<b>Formula:</b>	C5H8O2
<b>SMILES:</b>	CC=CCC(=O)O
<b>Mol. weight [g/mol]:</b>	100.12
<b>CAS:</b>	5204-64-8

## Physical Properties

Property code	Value	Unit	Source
chl	-2676.10 ± 2.10	kJ/mol	NIST Webbook
gf	-194.30	kJ/mol	Joback Method
hf	-294.12	kJ/mol	Joback Method
hfus	14.59	kJ/mol	Joback Method
hvap	50.11	kJ/mol	Joback Method
log10ws	-0.87		Crippen Method
logp	1.037		Crippen Method
mcvol	84.450	ml/mol	McGowan Method
pc	4498.26	kPa	Joback Method
tb	464.01	K	Joback Method
tc	644.72	K	Joback Method
tf	251.78	K	Joback Method
vc	0.321	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	166.69	J/molxK	464.01	Joback Method
cpg	173.98	J/molxK	494.13	Joback Method
cpg	180.90	J/molxK	524.25	Joback Method
cpg	187.47	J/molxK	554.36	Joback Method
cpg	193.71	J/molxK	584.48	Joback Method
cpg	199.64	J/molxK	614.60	Joback Method
cpg	205.26	J/molxK	644.72	Joback Method
dvisc	0.0276361	Paxs	251.78	Joback Method

dvisc	0.0072921	Paxs	287.15	Joback Method
dvisc	0.0025772	Paxs	322.52	Joback Method
dvisc	0.0011187	Paxs	357.89	Joback Method
dvisc	0.0005643	Paxs	393.27	Joback Method
dvisc	0.0003187	Paxs	428.64	Joback Method
dvisc	0.0001963	Paxs	464.01	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	368.00 ± 1.00	K	2.30	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.55457e+01
Coeff. B	-4.33606e+03
Coeff. C	-7.07400e+01
Temperature range (K), min.	354.92
Temperature range (K), max.	494.42

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5204648&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5204648&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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