

Diethylmalonic acid, heptadecyl 2-methylpentyl ester

Inchi: InChI=1S/C30H58O4/c1-6-10-11-12-13-14-15-16-17-18-19-20-21-22-23-25-33-28(31)30
InchiKey: UIGBPXCNJURDAV-UHFFFAOYSA-N
Formula: C30H58O4
SMILES: CCCCCCCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)OCC(C)CCC
Mol. weight [g/mol]: 482.78

Physical Properties

| Property code | Value | Unit | Source |
|---------------|----------|----------------------|----------------|
| gf | -265.72 | kJ/mol | Joback Method |
| hf | -1166.16 | kJ/mol | Joback Method |
| hfus | 68.09 | kJ/mol | Joback Method |
| hvap | 99.00 | kJ/mol | Joback Method |
| log10ws | -9.62 | | Crippen Method |
| logp | 9.187 | | Crippen Method |
| mvol | 448.440 | ml/mol | McGowan Method |
| pc | 631.61 | kPa | Joback Method |
| rinpol | 2962.00 | | NIST Webbook |
| rinpol | 2962.00 | | NIST Webbook |
| tb | 1034.71 | K | Joback Method |
| tc | 1288.59 | K | Joback Method |
| tf | 559.60 | K | Joback Method |
| vc | 1.746 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 1604.99 | J/molxK | 1034.71 | Joback Method |
| cpg | 1628.42 | J/molxK | 1077.02 | Joback Method |
| cpg | 1649.75 | J/molxK | 1119.34 | Joback Method |
| cpg | 1669.10 | J/molxK | 1161.65 | Joback Method |
| cpg | 1686.61 | J/molxK | 1203.97 | Joback Method |
| cpg | 1702.40 | J/molxK | 1246.28 | Joback Method |
| cpg | 1716.62 | J/molxK | 1288.59 | Joback Method |
| dvisc | 0.0002345 | Paxs | 559.60 | Joback Method |

| | | | | |
|-------|-----------|------|---------|---------------|
| dvisc | 0.0000935 | Paxs | 638.78 | Joback Method |
| dvisc | 0.0000457 | Paxs | 717.97 | Joback Method |
| dvisc | 0.0000257 | Paxs | 797.15 | Joback Method |
| dvisc | 0.0000161 | Paxs | 876.34 | Joback Method |
| dvisc | 0.0000109 | Paxs | 955.52 | Joback Method |
| dvisc | 0.0000078 | Paxs | 1034.71 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U369769&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

| | |
|----------------------------|---|
| cp_g: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log₁₀ws: | Log ₁₀ of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| m_{cvol}: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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