

# Suberoyl chloride

<b>Other names:</b>	Suberyl chloride Suberoyl dichloride Suberic acid dichloride Suberyl dichloride Octanedioyl dichloride
<b>Inchi:</b>	InChI=1S/C8H12Cl2O2/c9-7(11)5-3-1-2-4-6-8(10)12/h1-6H2
<b>InchiKey:</b>	PUIBKAHUQOOLSW-UHFFFAOYSA-N
<b>Formula:</b>	C8H12Cl2O2
<b>SMILES:</b>	O=C(Cl)CCCCCCC(=O)Cl
<b>Mol. weight [g/mol]:</b>	211.09
<b>CAS:</b>	10027-07-3

## Physical Properties

Property code	Value	Unit	Source
gf	-265.22	kJ/mol	Joback Method
hf	-465.09	kJ/mol	Joback Method
hfus	28.07	kJ/mol	Joback Method
hvap	55.66	kJ/mol	Joback Method
log10ws	-3.03		Crippen Method
logp	2.858		Crippen Method
mvol	151.200	ml/mol	McGowan Method
pc	2673.54	kPa	Joback Method
rinpol	1460.00		NIST Webbook
rinpol	1460.00		NIST Webbook
tb	565.04	K	Joback Method
tc	760.72	K	Joback Method
tf	339.62	K	Joback Method
vc	0.594	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	326.44	J/mol×K	565.04	Joback Method
cpg	373.86	J/mol×K	728.11	Joback Method

cpg	365.43	J/molxK	695.49	Joback Method
cpg	356.49	J/molxK	662.88	Joback Method
cpg	347.03	J/molxK	630.27	Joback Method
cpg	337.02	J/molxK	597.65	Joback Method
cpg	381.80	J/molxK	760.72	Joback Method
dvisc	0.0003317	Paxs	565.04	Joback Method
dvisc	0.0004214	Paxs	527.47	Joback Method
dvisc	0.0005553	Paxs	489.90	Joback Method
dvisc	0.0007662	Paxs	452.33	Joback Method
dvisc	0.0011207	Paxs	414.76	Joback Method
dvisc	0.0017682	Paxs	377.19	Joback Method
dvisc	0.0030858	Paxs	339.62	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10027073&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10027073&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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