

# Methyl nitrite

<b>Other names:</b>	Nitrous acid, methyl ester CH3ONO Methylester kyseliny dusite Methyl ester of nitrous acid
<b>Inchi:</b>	InChI=1S/CH3NO2/c1-4-2-3/h1H3
<b>InchiKey:</b>	BLLFVUPNHCTMSV-UHFFFAOYSA-N
<b>Formula:</b>	CH3NO2
<b>SMILES:</b>	CON=O
<b>Mol. weight [g/mol]:</b>	61.04
<b>CAS:</b>	624-91-9

## Physical Properties

Property code	Value	Unit	Source
affp	798.90	kJ/mol	NIST Webbook
basg	766.40	kJ/mol	NIST Webbook
chg	-752.00 ± 3.00	kJ/mol	NIST Webbook
hf	-65.44 ± 0.84	kJ/mol	NIST Webbook
hf	-62.50 ± 1.10	kJ/mol	NIST Webbook
hf	-70.00 ± 3.00	kJ/mol	NIST Webbook
hfl	-67.15 ± 0.84	kJ/mol	NIST Webbook
hfl	-85.10 ± 1.00	kJ/mol	NIST Webbook
hvap	22.60 ± 0.20	kJ/mol	NIST Webbook
ie	11.00	eV	NIST Webbook
ie	10.47 ± 0.01	eV	NIST Webbook
ie	10.44	eV	NIST Webbook
ie	10.38 ± 0.03	eV	NIST Webbook
ie	10.30	eV	NIST Webbook
ie	10.38 ± 0.03	eV	NIST Webbook
ie	10.38	eV	NIST Webbook
log10ws	-0.51		Crippen Method
logp	0.314		Crippen Method
mcvol	42.370	ml/mol	McGowan Method
pc	5495.11	kPa	Joback Method
rinpol	268.00		NIST Webbook
tb	308.10	K	Joback Method
tc	476.03	K	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hvapt	22.10	kJ/mol	245.50	NIST Webbook
hvapt	26.20	kJ/mol	189.50	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C624919&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C624919&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chg:</b>	Standard gas enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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