

D-Alanine, N-(3-anisoyl)-, nonyl ester

Inchi:	InChI=1S/C20H31NO4/c1-4-5-6-7-8-9-10-14-25-20(23)16(2)21-19(22)17-12-11-13-18(15)
InchiKey:	QIYHJCIVPXZPKG-UHFFFAOYSA-N
Formula:	C20H31NO4
SMILES:	CCCCCCCCCOC(=O)C(C)NC(=O)c1cccc(OC)c1
Mol. weight [g/mol]:	349.46

Physical Properties

Property code	Value	Unit	Source
gf	-160.59	kJ/mol	Joback Method
hf	-672.48	kJ/mol	Joback Method
hfus	48.36	kJ/mol	Joback Method
hvap	87.41	kJ/mol	Joback Method
log10ws	-5.51		Crippen Method
logp	4.107		Crippen Method
mvol	293.760	ml/mol	McGowan Method
pc	1362.64	kPa	Joback Method
rinpol	2737.00		NIST Webbook
rinpol	2737.00		NIST Webbook
tb	890.97	K	Joback Method
tc	1097.78	K	Joback Method
tf	536.08	K	Joback Method
vc	1.125	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	937.60	J/mol×K	890.97	Joback Method
cpg	952.96	J/mol×K	925.44	Joback Method
cpg	967.11	J/mol×K	959.91	Joback Method
cpg	980.05	J/mol×K	994.38	Joback Method
cpg	991.82	J/mol×K	1028.85	Joback Method
cpg	1002.44	J/mol×K	1063.31	Joback Method
cpg	1011.95	J/mol×K	1097.78	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U354048&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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