

# Aromadendra-4,10(14)-diene

<b>Inchi:</b>	InChI=1S/C15H22/c11-9-6-8-12-14(15(12,3)4)13-10(2)5-7-11(9)13/h11-12,14H,1,5-8H2,2
<b>InchiKey:</b>	DKAGYTWZAOJITC-UHFFFAOYSA-N
<b>Formula:</b>	C15H22
<b>SMILES:</b>	<chem>C=C1CCC2C(C3=C(C)CCC13)C2(C)C</chem>
<b>Mol. weight [g/mol]:</b>	202.34

## Physical Properties

Property code	Value	Unit	Source
gf	284.05	kJ/mol	Joback Method
hf	-32.87	kJ/mol	Joback Method
hfus	18.87	kJ/mol	Joback Method
hvap	49.38	kJ/mol	Joback Method
log10ws	-4.53		Crippen Method
logp	4.335		Crippen Method
mcvol	181.030	ml/mol	McGowan Method
pc	2102.27	kPa	Joback Method
rinpol	1442.00		NIST Webbook
rinpol	1442.00		NIST Webbook
tb	575.21	K	Joback Method
tc	794.86	K	Joback Method
tf	364.73	K	Joback Method
vc	0.698	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	484.12	J/molxK	575.21	Joback Method
cpg	505.10	J/molxK	611.82	Joback Method
cpg	524.76	J/molxK	648.43	Joback Method
cpg	543.26	J/molxK	685.04	Joback Method
cpg	560.79	J/molxK	721.65	Joback Method
cpg	577.52	J/molxK	758.25	Joback Method
cpg	593.62	J/molxK	794.86	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R427253&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R427253&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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