

# Dimethylmalonic acid, pentyl phenyl ester

<b>Inchi:</b>	InChI=1S/C16H22O4/c1-4-5-9-12-19-14(17)16(2,3)15(18)20-13-10-7-6-8-11-13/h6-8,10-
<b>InchiKey:</b>	XWVPCTNKPBVNHQ-UHFFFAOYSA-N
<b>Formula:</b>	C16H22O4
<b>SMILES:</b>	CCCCCOC(=O)C(C)(C)C(=O)Oc1ccccc1
<b>Mol. weight [g/mol]:</b>	278.34

## Physical Properties

Property code	Value	Unit	Source
gf	-268.75	kJ/mol	Joback Method
hf	-635.39	kJ/mol	Joback Method
hfus	29.40	kJ/mol	Joback Method
hvap	70.50	kJ/mol	Joback Method
log10ws	-3.75		Crippen Method
logp	3.352		Crippen Method
mvol	227.420	ml/mol	McGowan Method
pc	1870.79	kPa	Joback Method
rinpol	1816.00		NIST Webbook
rinpol	1816.00		NIST Webbook
tb	741.51	K	Joback Method
tc	951.19	K	Joback Method
tf	443.24	K	Joback Method
vc	0.861	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	651.40	J/molxK	741.51	Joback Method
cpg	667.00	J/molxK	776.46	Joback Method
cpg	681.51	J/molxK	811.40	Joback Method
cpg	694.98	J/molxK	846.35	Joback Method
cpg	707.44	J/molxK	881.30	Joback Method
cpg	718.93	J/molxK	916.25	Joback Method
cpg	729.50	J/molxK	951.19	Joback Method
dvisc	0.0009978	Paxs	443.24	Joback Method

dvisc	0.0005244	Paxs	492.95	Joback Method
dvisc	0.0003101	Paxs	542.66	Joback Method
dvisc	0.0002003	Paxs	592.38	Joback Method
dvisc	0.0001384	Paxs	642.09	Joback Method
dvisc	0.0001008	Paxs	691.80	Joback Method
dvisc	0.0000767	Paxs	741.51	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U361809&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U361809&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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