

# Salicylic acid, isopropyl ether, isopropyl ester

<b>Inchi:</b>	InChI=1S/C13H18O3/c1-9(2)15-12-8-6-5-7-11(12)13(14)16-10(3)4/h5-10H,1-4H3
<b>InchiKey:</b>	NUNWXYXSBPLFAO-UHFFFAOYSA-N
<b>Formula:</b>	C13H18O3
<b>SMILES:</b>	CC(C)OC(=O)c1ccccc1OC(C)C
<b>Mol. weight [g/mol]:</b>	222.28

## Physical Properties

Property code	Value	Unit	Source
gf	-182.44	kJ/mol	Joback Method
hf	-474.17	kJ/mol	Joback Method
hfus	20.01	kJ/mol	Joback Method
hvap	58.26	kJ/mol	Joback Method
log10ws	-3.73		Crippen Method
logp	3.039		Crippen Method
mcvol	183.580	ml/mol	McGowan Method
pc	2256.81	kPa	Joback Method
rinpol	1508.00		NIST Webbook
rinpol	1508.00		NIST Webbook
tb	626.33	K	Joback Method
tc	836.17	K	Joback Method
tf	339.60	K	Joback Method
vc	0.685	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	470.30	J/molxK	626.33	Joback Method
cpg	541.21	J/molxK	801.19	Joback Method
cpg	528.79	J/molxK	766.22	Joback Method
cpg	515.49	J/molxK	731.25	Joback Method
cpg	501.32	J/molxK	696.28	Joback Method
cpg	486.26	J/molxK	661.30	Joback Method
cpg	552.77	J/molxK	836.17	Joback Method
dvisc	0.0001138	Paxs	626.33	Joback Method

dvisc	0.0001498	Paxs	578.54	Joback Method
dvisc	0.0002072	Paxs	530.75	Joback Method
dvisc	0.0003054	Paxs	482.96	Joback Method
dvisc	0.0004904	Paxs	435.18	Joback Method
dvisc	0.0008851	Paxs	387.39	Joback Method
dvisc	0.0018861	Paxs	339.60	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U375433&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U375433&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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