

# Methyl radical, hydroxy-d1-

Inchi:	InChI=1S/CH3O/c1-2/h2H,1H2/i2D
InchiKey:	CBOIHMRHGLHBPB-VMNATFBRSA-N
Formula:	CH2DO
SMILES:	[CH2]O
Mol. weight [g/mol]:	32.04
CAS:	58456-46-5

## Physical Properties

Property code	Value	Unit	Source
gf	-126.90	kJ/mol	Joback Method
hf	-160.39	kJ/mol	Joback Method
hfus	4.12	kJ/mol	Joback Method
hvap	34.35	kJ/mol	Joback Method
ie	7.55 ± 0.01	eV	NIST Webbook
ie	7.55 ± 0.02	eV	NIST Webbook
log10ws	0.53		Crippen Method
logp	0.150		Crippen Method
mcvol	28.670	ml/mol	McGowan Method
pc	7206.25	kPa	Joback Method
tb	313.76	K	Joback Method
tc	471.75	K	Joback Method
tf	178.22	K	Joback Method
vc	0.102	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	38.13	J/mol×K	313.76	Joback Method
cpg	40.73	J/mol×K	340.09	Joback Method
cpg	43.12	J/mol×K	366.42	Joback Method
cpg	45.33	J/mol×K	392.76	Joback Method
cpg	47.35	J/mol×K	419.09	Joback Method
cpg	49.21	J/mol×K	445.42	Joback Method
cpg	50.92	J/mol×K	471.75	Joback Method

dvisc	0.0259356	Paxs	178.22	Joback Method
dvisc	0.0090382	Paxs	200.81	Joback Method
dvisc	0.0038981	Paxs	223.40	Joback Method
dvisc	0.0019620	Paxs	245.99	Joback Method
dvisc	0.0011084	Paxs	268.58	Joback Method
dvisc	0.0006842	Paxs	291.17	Joback Method
dvisc	0.0004527	Paxs	313.76	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C58456465&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C58456465&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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