

# 1-Tridecene

<b>Other names:</b>	1-C13H26 N-TRIDEC-1-ENE Tridec-1-ene Tridecene-1 «alpha»-Tridecene Â«alphaÂ»-Tridecene
<b>Inchi:</b>	InChI=1S/C13H26/c1-3-5-7-9-11-13-12-10-8-6-4-2/h3H,1,4-13H2,2H3
<b>InchiKey:</b>	VQOXUMQBYILCKR-UHFFFAOYSA-N
<b>Formula:</b>	C13H26
<b>SMILES:</b>	C=CCCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	182.35
<b>CAS:</b>	2437-56-1

## Physical Properties

Property code	Value	Unit	Source
af	0.5980		KDB
gf	146.40	kJ/mol	KDB
hf	-186.10	kJ/mol	KDB
hfus	28.15	kJ/mol	Joback Method
hvap	65.30	kJ/mol	NIST Webbook
ie	9.38	eV	NIST Webbook
ie	9.30 ± 0.05	eV	NIST Webbook
log10ws	-5.12		Crippen Method
logp	5.093		Crippen Method
mcvol	189.730	ml/mol	McGowan Method
pc	1700.00	kPa	KDB
rinpol	1291.30		NIST Webbook
rinpol	1293.00		NIST Webbook
rinpol	1292.00		NIST Webbook
rinpol	1287.00		NIST Webbook
rinpol	219.50		NIST Webbook
rinpol	1289.00		NIST Webbook
rinpol	1289.00		NIST Webbook
rinpol	1283.00		NIST Webbook
rinpol	1283.00		NIST Webbook
rinpol	1292.00		NIST Webbook
rinpol	1293.00		NIST Webbook

rinpol	1290.00	NIST Webbook
rinpol	1292.00	NIST Webbook
rinpol	1292.00	NIST Webbook
rinpol	1283.00	NIST Webbook
rinpol	1288.00	NIST Webbook
rinpol	1289.00	NIST Webbook
rinpol	1289.00	NIST Webbook
rinpol	1289.00	NIST Webbook
rinpol	1295.00	NIST Webbook
rinpol	1281.00	NIST Webbook
rinpol	1292.00	NIST Webbook
rinpol	1284.00	NIST Webbook
rinpol	1290.00	NIST Webbook
rinpol	1292.00	NIST Webbook
rinpol	1287.00	NIST Webbook
rinpol	1292.00	NIST Webbook
rinpol	1294.00	NIST Webbook
rinpol	1294.00	NIST Webbook
rinpol	1292.00	NIST Webbook
rinpol	1293.00	NIST Webbook
rinpol	1295.00	NIST Webbook
rinpol	1292.00	NIST Webbook
rinpol	1289.00	NIST Webbook
rinpol	1292.37	NIST Webbook
rinpol	1292.16	NIST Webbook
rinpol	1292.10	NIST Webbook
rinpol	1289.22	NIST Webbook
rinpol	1289.08	NIST Webbook
rinpol	1288.97	NIST Webbook
rinpol	1289.00	NIST Webbook
rinpol	1291.30	NIST Webbook
rinpol	1292.20	NIST Webbook
rinpol	1291.30	NIST Webbook
rinpol	1291.50	NIST Webbook
rinpol	1291.90	NIST Webbook
rinpol	1292.10	NIST Webbook
rinpol	1291.50	NIST Webbook
rinpol	1287.00	NIST Webbook
rinpol	1287.00	NIST Webbook
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rinpol	1292.20	NIST Webbook
rinpol	1291.30	NIST Webbook
rinpol	1291.50	NIST Webbook

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rinpol	1293.00	NIST Webbook
rinpol	1289.50	NIST Webbook
rinpol	1292.00	NIST Webbook
rinpol	1283.00	NIST Webbook
ripol	1343.00	NIST Webbook
ripol	1339.40	NIST Webbook
ripol	1340.90	NIST Webbook
ripol	1337.00	NIST Webbook
ripol	1344.10	NIST Webbook
ripol	1343.90	NIST Webbook
ripol	1373.00	NIST Webbook
ripol	1337.20	NIST Webbook
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ripol	1341.00	NIST Webbook
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ripol	1342.50	NIST Webbook

ripol	1342.00		NIST Webbook
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ripol	1331.90		NIST Webbook
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ripol	1340.80		NIST Webbook
ripol	1334.50		NIST Webbook
ripol	1331.60		NIST Webbook
ripol	1343.10		NIST Webbook
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ripol	1343.00		NIST Webbook
ripol	1342.00		NIST Webbook
ripol	1335.00		NIST Webbook
ripol	1343.00		NIST Webbook
ripol	1342.00		NIST Webbook
ripol	1342.00		NIST Webbook
ripol	1336.00		NIST Webbook
ripol	1341.00		NIST Webbook
tb	501.00 ± 5.00	K	NIST Webbook
tb	503.00 ± 8.00	K	NIST Webbook
tb	507.00 ± 2.00	K	NIST Webbook
tb	505.90	K	KDB
tb	505.00	K	NIST Webbook
tb	506.00	K	NIST Webbook
tc	674.00	K	KDB
tf	251.00 ± 2.00	K	NIST Webbook
tf	251.00 ± 0.50	K	NIST Webbook
tf	250.10	K	KDB
tf	250.03 ± 0.06	K	NIST Webbook
tf	250.05 ± 0.50	K	NIST Webbook
tf	250.04 ± 0.04	K	NIST Webbook
tf	250.08 ± 0.03	K	NIST Webbook

tf	250.04 ± 0.05	K	NIST Webbook
tf	250.08 ± 0.03	K	NIST Webbook
vc	0.745	m <sup>3</sup> /kmol	KDB
zc	0.2258480		KDB

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	436.90	J/mol×K	493.52	Joback Method
cpg	453.93	J/mol×K	520.69	Joback Method
cpg	470.29	J/mol×K	547.87	Joback Method
cpg	485.99	J/mol×K	575.04	Joback Method
cpg	501.06	J/mol×K	602.21	Joback Method
cpg	515.52	J/mol×K	629.38	Joback Method
cpg	529.38	J/mol×K	656.56	Joback Method
dvisc	0.0010013	Paxs	320.85	Joback Method
dvisc	0.0020407	Paxs	277.68	Joback Method
dvisc	0.0054051	Paxs	234.51	Joback Method
dvisc	0.0005817	Paxs	364.01	Joback Method
dvisc	0.0003792	Paxs	407.18	Joback Method
dvisc	0.0002683	Paxs	450.35	Joback Method
dvisc	0.0002017	Paxs	493.52	Joback Method
hvapt	44.98	kJ/mol	505.90	KDB
hvapt	53.90	kJ/mol	461.00	NIST Webbook
rho	766.00	kg/m <sup>3</sup>	293.00	KDB
srf	0.03	N/m	298.20	KDB

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.50003e+01
Coeff. B	-4.56047e+03
Coeff. C	-6.57330e+01
Temperature range (K), min.	375.70
Temperature range (K), max.	536.43

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	8.60269e+01
Coeff. B	-1.02545e+04
Coeff. C	-9.94660e+00
Coeff. D	3.11629e-06
Temperature range (K), min.	250.08
Temperature range (K), max.	675.00

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol348.mol">https://www.thermo.com/files/research/kdb/mol/mol348.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2437561&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2437561&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="https://www.chemed.com/doc/models/crippen_log10ws">https://www.chemed.com/doc/models/crippen_log10ws</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=348">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=348</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>The Yaws Handbook of Vapor Pressure: Infinite dilution activity coefficients, specific retention volumes and solvation thermodynamics of hydrocarbons in C7H158 branched alkane solvent:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> <a href="https://www.doi.org/10.1016/j.fluid.2006.07.015">https://www.doi.org/10.1016/j.fluid.2006.07.015</a>

## Legend

<b>af:</b>	Acentric Factor
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>h vapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rho:</b>	Liquid Density
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>srf:</b>	Surface Tension
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume
<b>zc:</b>	Critical Compressibility

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