

# Benzene, 2,4-dichloro-1-methoxy-

<b>Other names:</b>	Anisole, 2,4-dichloro- 1,5-Dichloro-2-methoxybenzene 2,4-Dichloro-1-methoxybenzene 2,4-Dichloroanisole Benzene, 1,3-dichloro-4-methoxy
<b>Inchi:</b>	InChI=1S/C7H6Cl2O/c1-10-7-3-2-5(8)4-6(7)9/h2-4H,1H3
<b>InchiKey:</b>	CICQUFBZCADHHX-UHFFFAOYSA-N
<b>Formula:</b>	C7H6Cl2O
<b>SMILES:</b>	COc1ccc(Cl)cc1Cl
<b>Mol. weight [g/mol]:</b>	177.03
<b>CAS:</b>	553-82-2

## Physical Properties

Property code	Value	Unit	Source
gf	-27.65	kJ/mol	Joback Method
hf	-137.92	kJ/mol	Joback Method
hfus	16.73	kJ/mol	Joback Method
hvap	45.96	kJ/mol	Joback Method
log10ws	-2.96		Crippen Method
logp	3.002		Crippen Method
mcvol	116.080	ml/mol	McGowan Method
pc	3543.08	kPa	Joback Method
rinpol	1263.00		NIST Webbook
rinpol	1299.90		NIST Webbook
rinpol	1249.00		NIST Webbook
rinpol	1250.00		NIST Webbook
rinpol	1265.00		NIST Webbook
rinpol	1269.00		NIST Webbook
rinpol	1256.00		NIST Webbook
rinpol	1282.00		NIST Webbook
rinpol	1265.00		NIST Webbook
rinpol	1272.00		NIST Webbook
rinpol	1253.00		NIST Webbook
ripol	1885.00		NIST Webbook
ripol	1905.00		NIST Webbook
ripol	1902.00		NIST Webbook
ripol	1869.00		NIST Webbook

ripol	1868.00		NIST Webbook
ripol	1857.00		NIST Webbook
tb	493.48	K	Joback Method
tc	720.45	K	Joback Method
tf	302.18	K	Joback Method
vc	0.435	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	206.95	J/mol×K	493.48	Joback Method
cpg	247.90	J/mol×K	682.62	Joback Method
cpg	240.65	J/mol×K	644.79	Joback Method
cpg	232.93	J/mol×K	606.96	Joback Method
cpg	224.74	J/mol×K	569.14	Joback Method
cpg	216.08	J/mol×K	531.31	Joback Method
cpg	254.69	J/mol×K	720.45	Joback Method
dvisc	0.0002335	Paxs	493.48	Joback Method
dvisc	0.0002827	Paxs	461.60	Joback Method
dvisc	0.0003521	Paxs	429.71	Joback Method
dvisc	0.0004541	Paxs	397.83	Joback Method
dvisc	0.0006124	Paxs	365.95	Joback Method
dvisc	0.0008743	Paxs	334.06	Joback Method
dvisc	0.0013457	Paxs	302.18	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C553822&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C553822&amp;Units=SI</a>

## Legend

**cpg:** Ideal gas heat capacity

<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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