

Chlortoluron

| | |
|-----------------------------|--|
| Other names: | 3-(3-Chlor-4-methylphenyl)-1,1-dimethylharnstoff 3-(3-Chloro-4-methylphenyl)-1,1-dimethylurea 3-(3-Chloro-p-tolyl)-1,1-dimethylurea C 2242 CGA 15646 Chlorotoluron Clortokem Dicuran Dicuran 500FL N,N-Dimethyl-N'-(3-chloro-4-methylphenyl)urea N-(3-Chloro-4-methylphenyl)-N',N'-dimethylurea Tolurex Urea, 3-(3-chloro-p-tolyl)-1,1-dimethyl- Urea, N'-(3-chloro-4-methylphenyl)-N,N-dimethyl- |
| Inchi: | InChI=1S/C10H13ClN2O/c1-7-4-5-8(6-9(7)11)12-10(14)13(2)3/h4-6H,1-3H3,(H,12,14) |
| InchiKey: | JXCGFZXSOMJFOA-UHFFFAOYSA-N |
| Formula: | C10H13ClN2O |
| SMILES: | <chem>Cc1ccc(NC(=O)N(C)C)cc1Cl</chem> |
| Mol. weight [g/mol]: | 212.68 |
| CAS: | 15545-48-9 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|--------|--------------------------------------|
| gf | 185.79 | kJ/mol | Joback Method |
| hf | -43.46 | kJ/mol | Joback Method |
| hfus | 28.84 | kJ/mol | Joback Method |
| hvap | 61.06 | kJ/mol | Joback Method |
| log10ws | -3.46 | | Aqueous Solubility Prediction Method |
| log10ws | -3.48 | | Estimated Solubility Method |
| logp | 2.742 | | Crippen Method |
| mcvol | 161.770 | ml/mol | McGowan Method |
| pc | 2953.69 | kPa | Joback Method |
| tb | 618.75 | K | Joback Method |
| tc | 836.56 | K | Joback Method |
| tf | 421.25 | K | Aqueous Solubility Prediction Method |

vc

0.596

m³/kmol

Joback Method

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 384.26 | J/mol×K | 618.75 | Joback Method |
| cpg | 397.18 | J/mol×K | 655.05 | Joback Method |
| cpg | 409.24 | J/mol×K | 691.35 | Joback Method |
| cpg | 420.49 | J/mol×K | 727.66 | Joback Method |
| cpg | 430.97 | J/mol×K | 763.96 | Joback Method |
| cpg | 440.71 | J/mol×K | 800.26 | Joback Method |
| cpg | 449.76 | J/mol×K | 836.56 | Joback Method |

Sources

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

Aqueous Solubility Prediction Method:

<http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

Estimated Solubility Method:

http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C15545489&Units=SI>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |

tf: Normal melting (fusion) point

vc: Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/57-236-7/Chlortoluron.pdf>

Generated by Cheméo on 2024-04-23 11:21:06.790082261 +0000 UTC m=+16160515.710659578.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.