

# 1,2,4,5-tetramethylcyclohexane, cis

<b>Inchi:</b>	InChI=1S/C10H20/c1-7-5-9(3)10(4)6-8(7)2/h7-10H,5-6H2,1-4H3/t7-,8+,9+,10-
<b>InchiKey:</b>	VWWAILZUSKHANH-FIRGSJFUSA-N
<b>Formula:</b>	C10H20
<b>SMILES:</b>	CC1CC(C)C(C)CC1C
<b>Mol. weight [g/mol]:</b>	140.27

## Physical Properties

Property code	Value	Unit	Source
gf	34.64	kJ/mol	Joback Method
hf	-256.43	kJ/mol	Joback Method
hfus	16.70	kJ/mol	Joback Method
hvap	37.36	kJ/mol	Joback Method
log10ws	-2.94		Crippen Method
logp	3.325		Crippen Method
mcvol	140.900	ml/mol	McGowan Method
pc	2329.27	kPa	Joback Method
rinpol	1054.00		NIST Webbook
rinpol	1047.00		NIST Webbook
rinpol	1053.00		NIST Webbook
rinpol	1054.00		NIST Webbook
rinpol	1055.00		NIST Webbook
rinpol	1050.00		NIST Webbook
rinpol	1047.00		NIST Webbook
rinpol	1056.00		NIST Webbook
tb	433.74	K	Joback Method
tc	629.39	K	Joback Method
tf	197.12	K	Joback Method
vc	0.525	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	299.06	J/molxK	433.74	Joback Method
cpg	319.44	J/molxK	466.35	Joback Method

cpg	338.99	J/mol×K	498.96	Joback Method
cpg	357.72	J/mol×K	531.57	Joback Method
cpg	375.63	J/mol×K	564.17	Joback Method
cpg	392.74	J/mol×K	596.78	Joback Method
cpg	409.05	J/mol×K	629.39	Joback Method
dvisc	0.0014826	Paxs	197.12	Joback Method
dvisc	0.0008678	Paxs	236.56	Joback Method
dvisc	0.0005920	Paxs	275.99	Joback Method
dvisc	0.0004444	Paxs	315.43	Joback Method
dvisc	0.0003555	Paxs	354.87	Joback Method
dvisc	0.0002974	Paxs	394.30	Joback Method
dvisc	0.0002570	Paxs	433.74	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R138844&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R138844&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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