

Benzene, 1-bromo-4-(1-methylethyl)-

Other names:	1-Bromo-4-isopropylbenzene 2-(4-Bromophenyl)propane 2-(p-Bromophenyl)propane 4-Bromocumene 4-Bromoisopropylbenzene 4-Isopropylbromobenzene Cumene, p-bromo- p-Bromocumene p-Bromoisopropylbenzene
Inchi:	InChI=1S/C9H11Br/c1-7(2)8-3-5-9(10)6-4-8/h3-7H,1-2H3
InchiKey:	MOZHUOIQYVYEPN-UHFFFAOYSA-N
Formula:	C9H11Br
SMILES:	CC(C)c1ccc(Br)cc1
Mol. weight [g/mol]:	199.09
CAS:	586-61-8

Physical Properties

Property code	Value	Unit	Source
gf	139.56	kJ/mol	Joback Method
hf	17.02	kJ/mol	Joback Method
hfus	14.48	kJ/mol	Joback Method
hvap	44.61	kJ/mol	Joback Method
log10ws	-3.82		Crippen Method
logp	3.572		Crippen Method
mcvol	131.410	ml/mol	McGowan Method
pc	3505.43	kPa	Joback Method
tb	491.90	K	NIST Webbook
tb	491.92 ± 0.30	K	NIST Webbook
tc	733.98	K	Joback Method
tf	250.80 ± 0.02	K	NIST Webbook
tf	250.69 ± 0.05	K	NIST Webbook
vc	0.487	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	313.04	J/molxK	695.44	Joback Method
cpg	322.39	J/molxK	733.98	Joback Method
cpg	254.97	J/molxK	502.70	Joback Method
cpg	268.23	J/molxK	541.25	Joback Method
cpg	280.63	J/molxK	579.79	Joback Method
cpg	292.20	J/molxK	618.34	Joback Method
cpg	302.99	J/molxK	656.89	Joback Method
dvisc	0.0002557	Paxs	502.70	Joback Method
dvisc	0.0003242	Paxs	464.74	Joback Method
dvisc	0.0028478	Paxs	274.93	Joback Method
dvisc	0.0014935	Paxs	312.89	Joback Method
dvisc	0.0009006	Paxs	350.85	Joback Method
dvisc	0.0005995	Paxs	388.81	Joback Method
dvisc	0.0004290	Paxs	426.78	Joback Method
hvapt	50.40	kJ/mol	458.00	NIST Webbook
hvapt	51.10	kJ/mol	427.50	NIST Webbook

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	370.70	K	0.70	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.41417e+01
Coeff. B	-3.94280e+03
Coeff. C	-7.78840e+01
Temperature range (K), min.	362.48
Temperature range (K), max.	524.40

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C586618&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure

Legend

cp_g:	Ideal gas heat capacity
d_{visc}:	Dynamic viscosity
g_f:	Standard Gibbs free energy of formation
h_f:	Enthalpy of formation at standard conditions
h_{fus}:	Enthalpy of fusion at standard conditions
h_{vap}:	Enthalpy of vaporization at standard conditions
h_{vapt}:	Enthalpy of vaporization at a given temperature
log_{10ws}:	Log ₁₀ of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
pc:	Critical Pressure
pv_{ap}:	Vapor pressure
tb:	Normal Boiling Point Temperature
tb_{rp}:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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