

# 2,5-Dimetoxycymene

<b>Inchi:</b>	InChI=1S/C11H16O2/c1-8(2)10-7-9(12-3)5-6-11(10)13-4/h5-8H,1-4H3
<b>InchiKey:</b>	UOMQSUCWUDELLOW-UHFFFAOYSA-N
<b>Formula:</b>	C11H16O2
<b>SMILES:</b>	COc1ccc(OC)c(C(C)C)c1
<b>Mol. weight [g/mol]:</b>	180.24

## Physical Properties

Property code	Value	Unit	Source
gf	-77.55	kJ/mol	Joback Method
hf	-326.50	kJ/mol	Joback Method
hfus	16.36	kJ/mol	Joback Method
hvap	48.11	kJ/mol	Joback Method
log10ws	-2.90		Crippen Method
logp	2.827		Crippen Method
mcvol	153.830	ml/mol	McGowan Method
pc	2502.50	kPa	Joback Method
rinpol	1399.00		NIST Webbook
rinpol	1399.00		NIST Webbook
ripol	1852.00		NIST Webbook
ripol	1852.00		NIST Webbook
tb	532.12	K	Joback Method
tc	737.45	K	Joback Method
tf	294.65	K	Joback Method
vc	0.574	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	351.77	J/molxK	532.12	Joback Method
cpg	366.81	J/molxK	566.34	Joback Method
cpg	381.19	J/molxK	600.56	Joback Method
cpg	394.92	J/molxK	634.78	Joback Method
cpg	407.99	J/molxK	669.00	Joback Method
cpg	420.40	J/molxK	703.22	Joback Method

cpg	432.15	J/molxK	737.45	Joback Method
dvisc	0.0014340	Paxs	294.65	Joback Method
dvisc	0.0007663	Paxs	334.23	Joback Method
dvisc	0.0004676	Paxs	373.81	Joback Method
dvisc	0.0003137	Paxs	413.38	Joback Method
dvisc	0.0002256	Paxs	452.96	Joback Method
dvisc	0.0001711	Paxs	492.54	Joback Method
dvisc	0.0001352	Paxs	532.12	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R518186&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R518186&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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