

2-Methyl-2-heptene

Other names:	2-Heptene, 2-methyl- 2-Methylhept-2-ene
Inchi:	InChI=1S/C8H16/c1-4-5-6-7-8(2)3/h7H,4-6H2,1-3H3
InchiKey:	WEPNJTDVIIKRIK-UHFFFAOYSA-N
Formula:	C8H16
SMILES:	CCCCC=C(C)C
Mol. weight [g/mol]:	112.21
CAS:	627-97-4

Physical Properties

Property code	Value	Unit	Source
gf	88.15	kJ/mol	Joback Method
hf	-101.02	kJ/mol	Joback Method
hfus	15.37	kJ/mol	Joback Method
hvap	39.70	kJ/mol	NIST Webbook
log10ws	-3.02		Crippen Method
logp	3.143		Crippen Method
mcvol	119.280	ml/mol	McGowan Method
pc	2687.42	kPa	Joback Method
rinpol	800.00		NIST Webbook
rinpol	791.00		NIST Webbook
rinpol	789.50		NIST Webbook
rinpol	800.00		NIST Webbook
rinpol	800.00		NIST Webbook
rinpol	786.00		NIST Webbook
rinpol	770.00		NIST Webbook
rinpol	796.00		NIST Webbook
rinpol	770.00		NIST Webbook
rinpol	789.50		NIST Webbook
rinpol	786.00		NIST Webbook
rinpol	787.00		NIST Webbook
rinpol	789.60		NIST Webbook
rinpol	800.20		NIST Webbook
rinpol	789.50		NIST Webbook
rinpol	786.00		NIST Webbook
tb	395.75 ± 0.50	K	NIST Webbook
tb	395.85 ± 0.40	K	NIST Webbook

tb	398.35 ± 0.60	K	NIST Webbook
tb	394.85 ± 0.70	K	NIST Webbook
tb	394.65 ± 1.00	K	NIST Webbook
tb	393.65 ± 2.00	K	NIST Webbook
tb	394.75 ± 0.70	K	NIST Webbook
tb	395.80	K	NIST Webbook
tb	395.75 ± 0.60	K	NIST Webbook
tb	395.77 ± 0.20	K	NIST Webbook
tb	396.02 ± 0.20	K	NIST Webbook
tb	395.83 ± 0.20	K	NIST Webbook
tc	568.90	K	Gas-Liquid Critical Temperatures of Some Alkenes, Amines, and Cyclic Hydrocarbons
tf	151.00 ± 0.02	K	NIST Webbook
tf	150.98 ± 0.06	K	NIST Webbook
tf	150.98 ± 0.06	K	NIST Webbook
vc	0.465	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	278.15	J/mol×K	532.54	Joback Method
cpg	218.03	J/mol×K	386.48	Joback Method
cpg	231.15	J/mol×K	415.69	Joback Method
cpg	243.70	J/mol×K	444.90	Joback Method
cpg	255.70	J/mol×K	474.11	Joback Method
cpg	267.18	J/mol×K	503.32	Joback Method
cpg	288.64	J/mol×K	561.75	Joback Method
hvapt	41.20	kJ/mol	326.50	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.42166e+01
Coeff. B	-3.33544e+03
Coeff. C	-4.84960e+01

Temperature range (K), min.	287.96
Temperature range (K), max.	423.05

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	9.49092e+01
Coeff. B	-7.89919e+03
Coeff. C	-1.19986e+01
Coeff. D	9.12422e-06
Temperature range (K), min.	288.15
Temperature range (K), max.	423.15

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C627974&Units=SI
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Gas-Liquid Critical Temperatures of Some Alkenes, Amines, and Cyclic Hydrocarbons:	https://www.doi.org/10.1021/je0341357
KDB Vapor Pressure Data:	https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=261
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
KDB:	https://www.thermo.com/files/research/kdb/mol/mol261.mol

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure

pvap:	Vapor pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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