

# Ethane, 1-bromo-2-chloro-

<b>Other names:</b>	1,2-Bromochloroethane 1,2-Chlorobromoethane 1-Bromo-2-chloroethane 1-Chloro-2-bromoethane 1-bromo-2-chloroethyne 2-Bromo-1-chloroethane 2-Bromoethyl chloride 2-Chloroethyl bromide CH <sub>2</sub> BrCH <sub>2</sub> Cl Ethylene chlorobromide NSC 60186 s-Chlorobromoethane sym-Chlorobromoethane «beta»-Chloroethyl bromide Â«betaÂ»-Chloroethyl bromide
<b>Inchi:</b>	InChI=1S/C2H4BrCl/c3-1-2-4/h1-2H2
<b>InchiKey:</b>	IBYHHJPAARCAIE-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>2</sub> H <sub>4</sub> BrCl
<b>SMILES:</b>	CICCB
<b>Mol. weight [g/mol]:</b>	143.41
<b>CAS:</b>	107-04-0

## Physical Properties

Property code	Value	Unit	Source
gf	-31.65	kJ/mol	Joback Method
hf	-74.02	kJ/mol	Joback Method
hfus	10.42	kJ/mol	Joback Method
hvap	38.13 ± 0.04	kJ/mol	NIST Webbook
ie	10.55	eV	NIST Webbook
ie	10.63 ± 0.03	eV	NIST Webbook
ie	10.50 ± 0.10	eV	NIST Webbook
ie	10.70 ± 0.10	eV	NIST Webbook
ie	10.65 ± 0.01	eV	NIST Webbook
ie	10.52	eV	NIST Webbook
ie	10.57 ± 0.05	eV	NIST Webbook
ie	10.63	eV	NIST Webbook

log10ws	-1.32		Aqueous Solubility Prediction Method
log10ws	-1.32		Estimated Solubility Method
logp	1.620		Crippen Method
mcvol	68.780	ml/mol	McGowan Method
pc	5266.25	kPa	Joback Method
rinpol	704.00		NIST Webbook
rinpol	723.00		NIST Webbook
rinpol	716.00		NIST Webbook
rinpol	725.20		NIST Webbook
rinpol	688.00		NIST Webbook
rinpol	693.00		NIST Webbook
rinpol	698.00		NIST Webbook
rinpol	729.00		NIST Webbook
rinpol	681.00		NIST Webbook
rinpol	708.00		NIST Webbook
ripol	1174.97		NIST Webbook
ripol	1186.26		NIST Webbook
ripol	1186.26		NIST Webbook
ripol	1192.67		NIST Webbook
tb	348.75	K	Joback Method
tc	545.22	K	Joback Method
tf	256.40	K	NIST Webbook
tf	198.80 ± 0.40	K	NIST Webbook
tf	256.50 ± 0.20	K	NIST Webbook
tf	255.92	K	Aqueous Solubility Prediction Method
vc	0.259	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	86.29	J/mol×K	348.75	Joback Method
cpg	94.69	J/mol×K	414.24	Joback Method
cpg	98.54	J/mol×K	446.99	Joback Method
cpg	102.17	J/mol×K	479.73	Joback Method
cpg	105.59	J/mol×K	512.48	Joback Method
cpg	108.82	J/mol×K	545.22	Joback Method
cpg	90.61	J/mol×K	381.50	Joback Method

cpl	136.77	J/molxK	295.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	136.84	J/molxK	297.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	136.89	J/molxK	298.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	136.92	J/molxK	298.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	136.99	J/molxK	300.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K

cpl	137.07	J/mol×K	301.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	137.15	J/mol×K	303.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	137.23	J/mol×K	304.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	137.31	J/mol×K	306.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	137.39	J/mol×K	307.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K

cpl	137.48	J/molxK	309.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	137.57	J/molxK	310.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	137.66	J/molxK	312.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	137.75	J/molxK	313.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	137.84	J/molxK	315.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K

cpl	137.93	J/mol×K	316.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	138.03	J/mol×K	318.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	138.13	J/mol×K	319.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	138.22	J/mol×K	321.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	138.33	J/mol×K	322.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K

cpl	138.43	J/molxK	324.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	138.53	J/molxK	325.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	138.64	J/molxK	327.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	138.74	J/molxK	328.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	138.85	J/molxK	330.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K

cpl	138.96	J/mol×K	331.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	139.07	J/mol×K	333.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	139.19	J/mol×K	334.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	139.30	J/mol×K	336.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	139.42	J/mol×K	337.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K



cpl	139.54	J/molxK	339.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	139.66	J/molxK	340.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	139.78	J/molxK	342.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	139.90	J/molxK	343.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	140.03	J/molxK	345.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K

cpl	140.16	J/mol×K	346.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	140.29	J/mol×K	348.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	140.42	J/mol×K	349.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	140.55	J/mol×K	351.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	140.68	J/mol×K	352.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K

cpl	140.82	J/mol×K	354.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	136.70	J/mol×K	294.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	130.10	J/mol×K	300.00	NIST Webbook
cpl	136.63	J/mol×K	292.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	136.56	J/mol×K	291.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	136.50	J/mol×K	289.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K

cpl	136.43	J/molxK	288.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	136.37	J/molxK	286.65	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	140.91	J/molxK	355.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
cpl	136.31	J/molxK	285.15	Heat Capacity of alpha,omega-Bromochloroalkanes and ?,?-Dibromoalkanes: Their Dependence on the Hydrocarbon Chain Length and Temperature (285.15 to 355.15) K
dvisc	0.0004315	Paxs	348.75	Joback Method
dvisc	0.0005304	Paxs	324.30	Joback Method
dvisc	0.0031520	Paxs	202.02	Joback Method
dvisc	0.0018920	Paxs	226.48	Joback Method
dvisc	0.0012545	Paxs	250.93	Joback Method
dvisc	0.0008948	Paxs	275.38	Joback Method
dvisc	0.0006744	Paxs	299.84	Joback Method
hfust	3.10	kJ/mol	182.00	NIST Webbook
hfust	9.62	kJ/mol	256.40	NIST Webbook
hfust	9.62	kJ/mol	256.40	NIST Webbook
hvapt	39.50	kJ/mol	311.50	NIST Webbook
hvapt	36.40 ± 0.10	kJ/mol	338.00	NIST Webbook
hvapt	36.60 ± 0.10	kJ/mol	330.00	NIST Webbook
hvapt	36.90 ± 0.10	kJ/mol	323.00	NIST Webbook

hvapt	37.30 ± 0.10	kJ/mol	315.00	NIST Webbook
hvapt	38.06	kJ/mol	379.80	NIST Webbook
hvapt	37.60 ± 0.10	kJ/mol	308.00	NIST Webbook
pvap	8.91	kPa	313.15	Isothermal Vapor-Liquid Equilibria of ethyl acetate + dibromomethane, or + bromochloromethane or + 1,2-dichloroethane or +1-bromo-2-chloroethane at T = 313.15 K
rho1	1727.01	kg/m <sup>3</sup>	298.15	(Vapor + liquid) equilibria for the binary mixtures (1-propanol + dibromomethane, or + bromochloromethane, or + 1,2-dichloroethane or +1-bromo-2-chloroethane) at T = 313.15 K.
sfust	37.53	J/mol×K	256.40	NIST Webbook
sfust	17.15	J/mol×K	182.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.31363e+01
Coeff. B	-2.60209e+03
Coeff. C	-7.46670e+01
Temperature range (K), min.	272.75
Temperature range (K), max.	407.21

## Sources

- Temperature and Pressure Dependence of the Volumetric Properties of Binary Liquid Mixtures Containing Dihalalkanes: The Yaws Handbook of Vapor Pressure: <https://www.doi.org/10.1007/s10765-005-5570-x>
- Joback Method: [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)
- The Yaws Handbook of Vapor Pressure: <https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

