

# Diethylmalonic acid, 2-fluorophenyl nonyl ester

Inchi:

InChI=1S/C22H33FO4/c1-4-7-8-9-10-11-14-17-26-20(24)22(5-2,6-3)21(25)27-19-16-13-

InchiKey:

NHYNGMMIUWUANQ-UHFFFAOYSA-N

Formula:

C22H33FO4

SMILES:

CCCCCCCCCOC(=O)C(CC)(CC)C(=O)Oc1ccccc1F

Mol. weight [g/mol]:

380.49

## Physical Properties

Property code	Value	Unit	Source
gf	-422.67	kJ/mol	Joback Method
hf	-966.81	kJ/mol	Joback Method
hfus	47.63	kJ/mol	Joback Method
hvap	83.70	kJ/mol	Joback Method
log10ws	-6.60		Crippen Method
logp	5.831		Crippen Method
mcvol	313.730	ml/mol	McGowan Method
pc	1150.65	kPa	Joback Method
rinpol	2388.00		NIST Webbook
tb	883.04	K	Joback Method
tc	1086.27	K	Joback Method
tf	523.97	K	Joback Method
vc	1.214	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1005.03	J/molxK	883.04	Joback Method
cpg	1021.31	J/molxK	916.91	Joback Method
cpg	1036.40	J/molxK	950.78	Joback Method
cpg	1050.37	J/molxK	984.65	Joback Method
cpg	1063.26	J/molxK	1018.52	Joback Method
cpg	1075.10	J/molxK	1052.40	Joback Method
cpg	1085.96	J/molxK	1086.27	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370133&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370133&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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