

# Methyl 4-chloro-3-methylbenzoate

<b>Other names:</b>	Benzoic acid, 4-chloro-3-methyl-, methyl ester
<b>Inchi:</b>	InChI=1S/C9H9ClO2/c1-6-5-7(9(11)12-2)3-4-8(6)10/h3-5H,1-2H3
<b>InchiKey:</b>	QOTGNXMPQNGYDM-UHFFFAOYSA-N
<b>Formula:</b>	C9H9ClO2
<b>SMILES:</b>	<chem>COC(=O)c1ccc(Cl)c(C)c1</chem>
<b>Mol. weight [g/mol]:</b>	184.62
<b>CAS:</b>	91367-05-4

## Physical Properties

Property code	Value	Unit	Source
gf	-127.80	kJ/mol	Joback Method
hf	-276.04	kJ/mol	Joback Method
hfus	19.31	kJ/mol	Joback Method
hvap	52.77	kJ/mol	Joback Method
log10ws	-2.87		Crippen Method
logp	2.435		Crippen Method
mcvol	133.590	ml/mol	McGowan Method
pc	3184.73	kPa	Joback Method
tb	555.68	K	Joback Method
tc	778.87	K	Joback Method
tf	344.73	K	Joback Method
vc	0.504	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	280.08	J/molxK	555.68	Joback Method
cpg	291.31	J/molxK	592.88	Joback Method
cpg	301.92	J/molxK	630.08	Joback Method
cpg	311.91	J/molxK	667.27	Joback Method
cpg	321.30	J/molxK	704.47	Joback Method
cpg	330.08	J/molxK	741.67	Joback Method
cpg	338.26	J/molxK	778.87	Joback Method
dvisc	0.0013360	Paxs	344.73	Joback Method

dvisc	0.0008629	Paxs	379.89	Joback Method
dvisc	0.0006002	Paxs	415.05	Joback Method
dvisc	0.0004418	Paxs	450.20	Joback Method
dvisc	0.0003399	Paxs	485.36	Joback Method
dvisc	0.0002710	Paxs	520.52	Joback Method
dvisc	0.0002223	Paxs	555.68	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C91367054&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C91367054&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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