

# p-n-Hexyloxybenzylideneaniline

<b>Inchi:</b>	InChI=1S/C19H23NO/c1-2-3-4-8-15-21-19-13-11-17(12-14-19)16-20-18-9-6-5-7-10-18/h
<b>InchiKey:</b>	IBVAICKBWALIQR-UHFFFAOYSA-N
<b>Formula:</b>	C19H23NO
<b>SMILES:</b>	CCCCCOc1ccc(C=Nc2ccccc2)cc1
<b>Mol. weight [g/mol]:</b>	281.39
<b>CAS:</b>	5219-49-8

## Physical Properties

Property code	Value	Unit	Source
hf	-23.90	kJ/mol	Joback Method
hvap	68.83	kJ/mol	Joback Method
log10ws	-5.55		Crippen Method
logp	5.396		Crippen Method
mvol	242.600	ml/mol	McGowan Method
pc	1578.46	kPa	Joback Method
ss	424.16	J/molxK	NIST Webbook
tb	791.56	K	Joback Method
tc	1020.05	K	Joback Method
tt	321.63 ± 0.02	K	NIST Webbook
tt	321.63 ± 0.02	K	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cps	404.52	J/molxK	298.15	NIST Webbook
hfust	30.91	kJ/mol	321.60	NIST Webbook

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C5219498&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>  
**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)  
**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

**cps:** Solid phase heat capacity  
**hf:** Enthalpy of formation at standard conditions  
**hfust:** Enthalpy of fusion at a given temperature  
**hvap:** Enthalpy of vaporization at standard conditions  
**log10ws:** Log10 of Water solubility in mol/l  
**logp:** Octanol/Water partition coefficient  
**mcvol:** McGowan's characteristic volume  
**pc:** Critical Pressure  
**ss:** Solid phase molar entropy at standard conditions  
**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tt:** Triple Point Temperature

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