

# 2-Propenoic acid, 2-methyl-, 2-hydroxypropyl ester

Other names:	1,2-Propanediol, 1-methacrylate 2-hydroxypropyl 2-methyl-2-propenoate 2-hydroxypropyl methacrylate methacrylic acid, 2-hydroxypropyl ester «beta»-Hydroxypropyl methacrylate
Inchi:	InChI=1S/C7H12O3/c1-5(2)7(9)10-4-6(3)8/h6,8H,1,4H2,2-3H3
InchiKey:	VHSJLMUCYSAUQU-UHFFFAOYSA-N
Formula:	C7H12O3
SMILES:	C=C(C)C(=O)OCC(C)O
Mol. weight [g/mol]:	144.17
CAS:	923-26-2

## Physical Properties

Property code	Value	Unit	Source
gf	-285.83	kJ/mol	Joback Method
hf	-474.48	kJ/mol	Joback Method
hfus	14.65	kJ/mol	Joback Method
hvap	56.03	kJ/mol	Joback Method
log10ws	-0.84		Crippen Method
logp	0.486		Crippen Method
mcvol	118.500	ml/mol	McGowan Method
pc	3472.45	kPa	Joback Method
tb	524.15	K	Joback Method
tc	703.12	K	Joback Method
tf	270.91	K	Joback Method
vc	0.447	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	270.15	J/mol×K	524.15	Joback Method
cpg	279.73	J/mol×K	553.98	Joback Method
cpg	288.90	J/mol×K	583.81	Joback Method
cpg	297.68	J/mol×K	613.63	Joback Method

cpg	306.08	J/mol×K	643.46	Joback Method
cpg	314.09	J/mol×K	673.29	Joback Method
cpg	321.72	J/mol×K	703.12	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C923262&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C923262&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>High pressure phase behaviour of the binary mixture for the 2-hydroxyethyl methacrylate, 2-hydroxypropyl acrylate, and 2-hydroxypropyl methacrylate in supercritical carbon dioxide:</b>	<a href="https://www.doi.org/10.1016/j.jct.2006.11.010">https://www.doi.org/10.1016/j.jct.2006.11.010</a>
	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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